

Density Worksheet

Physical Science

$D = m/V$

Densities of Common Substances @ 20°C				
Substance	Density (g/cm ³)		Substance	Density (g/cm ³)
Oxygen	0.00133		Aluminum	2.70
Hydrogen	0.000084		Iron	7.87
Ethanol	0.785		Copper	8.96
Benzene	0.880		Silver	10.5
Water	1.000		Lead	11.34
Magnesium	1.74		Mercury	13.6
Salt (sodium chloride)	2.16		Gold	19.32

- The ratio of an object's mass to its _____ is called the *density* of the object.
- A kilogram of lead occupies a much smaller volume than a kilogram of water, because _____ has a much higher *density*.
- For the masses and volumes indicated, calculate the **density** in grams per cubic centimeters.
 - mass = 453 g; volume = 225 cm³
 - mass = 5.0 g; volume = 10.0 cm³
 - mass = 26.1 g; volume = 2.0 mL
- If 89.2 mL of a liquid has a mass of 75.2 g, calculate the liquid's density.
- A cube of metal weighs 1450 g and displaces 542 mL of water when immersed. Calculate the density of the metal.

Physical Science Worksheet Answers

G Orfield



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Table of Contents Physical Science Worksheet Answers

1. Understanding the eBook Physical Science Worksheet Answers
 - The Rise of Digital Reading Physical Science Worksheet Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Physical Science Worksheet Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Physical Science Worksheet Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Physical Science Worksheet Answers
 - Personalized Recommendations
 - Physical Science Worksheet Answers User Reviews and Ratings
 - Physical Science Worksheet Answers and Bestseller Lists
5. Accessing Physical Science Worksheet Answers Free and Paid eBooks
 - Physical Science Worksheet Answers Public Domain eBooks
 - Physical Science Worksheet Answers eBook Subscription Services
 - Physical Science Worksheet Answers Budget-Friendly Options
6. Navigating Physical Science Worksheet Answers eBook Formats

- ePub, PDF, MOBI, and More
 - Physical Science Worksheet Answers Compatibility with Devices
 - Physical Science Worksheet Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Physical Science Worksheet Answers
 - Highlighting and Note-Taking Physical Science Worksheet Answers
 - Interactive Elements Physical Science Worksheet Answers
 8. Staying Engaged with Physical Science Worksheet Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Physical Science Worksheet Answers
 9. Balancing eBooks and Physical Books Physical Science Worksheet Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Physical Science Worksheet Answers
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Physical Science Worksheet Answers
 - Setting Reading Goals Physical Science Worksheet Answers
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Physical Science Worksheet Answers
 - Fact-Checking eBook Content of Physical Science Worksheet Answers
 - Distinguishing Credible Sources
 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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