



**PHYSICAL SCIENCES (CHEMISTRY)
GRADE 11 TERM 1 FORMAL EXPERIMENT - PART 2
VERIFICATION OF BOYLE'S LAW
MARKING GUIDELINES**

- 1.1 • Aim stated correctly. ✓
e.g. To determine the relationship between the volume of a fixed amount of gas and the pressure on the gas when temperature remains constant.

(1)

- 1.2 • All apparatus listed ✓

1. Boyle's Apparatus
2. Pump

- All steps of the method correct and logically stated. ✓✓ correct tense used. ✓✓

e.g.

1. The tap on the Boyle's Law apparatus was opened and then the pump was connected.
2. The apparatus was pumped to the pressure.
3. The oil was allowed to settle for a few minutes.
4. A table was used to record the volume reading on the tube and the corresponding pressure reading on the Bourdon pressure gauge.
5. The air tap was slightly opened and quickly closed to let a little air escape from the system. The oil was allowed to settle for a few minutes and then the pressure and volume readings were taken.
6. 5 different readings were recorded in the table.

(5)

2. Observations and Recording of results:

Learners will use results obtained during experiment to complete the table below.

(6)

Volume (V) cm ³	Pressure (p) kPa		pV (°)

Marking guidelines:

- ✓ for correct values of the volume
- ✓ for correct values of the pressure
- ✓✓ for correct values of

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A Gutmann



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Table of Contents Physical Science Term 3 Practical Memo Grade 12

1. Understanding the eBook Physical Science Term 3 Practical Memo Grade 12
 - The Rise of Digital Reading Physical Science Term 3 Practical Memo Grade 12
 - Advantages of eBooks Over Traditional Books
2. Identifying Physical Science Term 3 Practical Memo Grade 12
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Physical Science Term 3 Practical Memo Grade 12
 - User-Friendly Interface
4. Exploring eBook Recommendations from Physical Science Term 3 Practical Memo Grade 12
 - Personalized Recommendations
 - Physical Science Term 3 Practical Memo Grade 12 User Reviews and Ratings
 - Physical Science Term 3 Practical Memo Grade 12 and Bestseller Lists
5. Accessing Physical Science Term 3 Practical Memo Grade 12 Free and Paid eBooks

- Physical Science Term 3 Practical Memo Grade 12 Public Domain eBooks
 - Physical Science Term 3 Practical Memo Grade 12 eBook Subscription Services
 - Physical Science Term 3 Practical Memo Grade 12 Budget-Friendly Options
6. Navigating Physical Science Term 3 Practical Memo Grade 12 eBook Formats
- ePub, PDF, MOBI, and More
 - Physical Science Term 3 Practical Memo Grade 12 Compatibility with Devices
 - Physical Science Term 3 Practical Memo Grade 12 Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of Physical Science Term 3 Practical Memo Grade 12
 - Highlighting and Note-Taking Physical Science Term 3 Practical Memo Grade 12
 - Interactive Elements Physical Science Term 3 Practical Memo Grade 12
8. Staying Engaged with Physical Science Term 3 Practical Memo Grade 12
- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Physical Science Term 3 Practical Memo Grade 12
9. Balancing eBooks and Physical Books Physical Science Term 3 Practical Memo Grade 12
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection Physical Science Term 3 Practical Memo Grade 12
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Physical Science Term 3 Practical Memo Grade 12
- Setting Reading Goals Physical Science Term 3 Practical Memo Grade 12
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Physical Science Term 3 Practical Memo Grade 12
- Fact-Checking eBook Content of Physical Science Term 3 Practical Memo Grade 12
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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