

Name _____

Physical Science Chemical Bonding Study Guide

A. Covalent & Ionic Compounds/Bonds

1. Which ion always forms with a charge of 1? _____ (anion)
Which cation is always a 2+ charge? _____ (cation)
2. Are elements in the halogen group less _____ valence electrons, and tends to _____ 1 more to have 8. This is because of the _____ rule.
3. Are _____ found in the alkaline elements especially charged ions.
PCl₃ is charged with a _____
4. On a molecular level, atoms _____ molecules, formed by an
ionic bond, atoms _____ molecules.
5. Elements group 16's also form ions charged negatively by a constant bond of 2.
_____ and others group 17's charged negatively by
two, and group 13's is charged, in fact, with a constant of negative charged by
three with the _____
6. Are atoms with more than 4 _____ valence electrons. This usually indicates an ionic
compound. But there are 3 exceptions especially in metals.
7. Atoms with 1, 2, or 3 _____ or 4 _____ by forming ionic
bonds.

B. Formulas for Ionic Compounds – Fill in the blank below

Write the correct electron configuration of	Magnesium	Fluorine	Phosphorus	Lithium

3. Write in the chemical formulas for each of the following ions:
compounds of

Think your way with these steps: Determine valence, and use arrows to show the transfer of
electrons. Use your periodic table to find their oxidation numbers. They will combine so that
their oxidation numbers cancel out to zero. Use the criss cross method if their
oxidation numbers do not cancel.

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Appropriations,United States. Congress. House. Committee on Appropriations. Subcommittee on Independent Offices,1965

Annual Report for Fiscal Year ... National Science Foundation (U.S.),1963 **Chemistry, Student Study Guide**

James E. Brady,Fred Senese,2008-01-28 The image on the front cover depicts a carbon nanotube emerging from a glowing plasma of hydrogen and carbon as it forms around particles of a metal catalyst Carbon nanotubes are a recently discovered allotrope of carbon Three other allotropes of carbon buckyballs graphite and diamond are illustrated at the left as is the molecule methane CH₄ from which nanotubes and buckyballs can be made The element carbon forms an amazing number of compounds with structures that follow from simple methane found in natural gas to the complex macromolecules that serve as the basis of life on our planet The study of chemistry also follows from the simple to the more complex and the strength of this text is that it enables students with varied backgrounds to proceed together to significant levels of achievement

INNOVATIVE SCIENCE TEACHING, FOURTH EDITION MOHAN, RADHA,2019-09-01 Science teaching has evolved as a blend of conventional methods and modern aids owing to the changing needs and techniques of education with an objective to develop scientific attitude among the students This Fourth Edition of Innovative Science Teaching aims to strike balance between modern teaching methods and time tested theories FEATURES OF THE FOURTH EDITION Chapters 3 8 and 13 have been thoroughly revised and updated in the light of advancements of application of technology in teaching Chapter 13 New Technology to Promote Learning has been expanded to include the impact of technology on teaching and learning E learning materials and website addresses relevant to science teaching have been updated All chapters have been revised and extensive coverage of all aspects of modern teaching has been included This edition of Innovative Science Teaching is designed for the undergraduate and postgraduate students of Education specializing in science teaching It can also prove useful as a reference book for administrators researchers and teacher trainers TARGET AUDIENCE B Ed specialization in Science Teaching M Ed specialization in Science Teaching Diploma Courses in Education **Resources for Teaching Middle School Science** Smithsonian Institution,National Academy of Engineering,National Science Resources

Center of the National Academy of Sciences, Institute of Medicine, 1998-04-30 With age appropriate inquiry centered curriculum materials and sound teaching practices middle school science can capture the interest and energy of adolescent students and expand their understanding of the world around them Resources for Teaching Middle School Science developed by the National Science Resources Center NSRC is a valuable tool for identifying and selecting effective science curriculum materials that will engage students in grades 6 through 8 The volume describes more than 400 curriculum titles that are aligned with the National Science Education Standards This completely new guide follows on the success of Resources for Teaching Elementary School Science the first in the NSRC series of annotated guides to hands on inquiry centered curriculum materials and other resources for science teachers The curriculum materials in the new guide are grouped in five chapters by scientific area Physical Science Life Science Environmental Science Earth and Space Science and Multidisciplinary and Applied Science They are also grouped by type core materials supplementary units and science activity books Each annotation of curriculum material includes a recommended grade level a description of the activities involved and of what students can be expected to learn a list of accompanying materials a reading level and ordering information The curriculum materials included in this book were selected by panels of teachers and scientists using evaluation criteria developed for the guide The criteria reflect and incorporate goals and principles of the National Science Education Standards The annotations designate the specific content standards on which these curriculum pieces focus In addition to the curriculum chapters the guide contains six chapters of diverse resources that are directly relevant to middle school science Among these is a chapter on educational software and multimedia programs chapters on books about science and teaching directories and guides to science trade books and periodicals for teachers and students Another section features institutional resources One chapter lists about 600 science centers museums and zoos where teachers can take middle school students for interactive science experiences Another chapter describes nearly 140 professional associations and U S government agencies that offer resources and assistance Authoritative extensive and thoroughly indexed and the only guide of its kind Resources for Teaching Middle School Science will be the most used book on the shelf for science teachers school administrators teacher trainers science curriculum specialists advocates of hands on science teaching and concerned parents

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