

#### MERCY JUNIOR COLLEGE Grade 11 Physical Science Midterm Exam

NAME:	
YEAR LEVEL & SECTION:	
TEACHER'S NAME:	
PROCTOR'S NAME:	

I. MULTIPLE CHOICE

INSTRUCTION: Encircle the letter of the correct answer. (1 POINT EACH)

It is a subatomic particle, symbol e- or

β-, with a negative elementary electric charge.

- A. Newtron
- 8. Electron
- C. Proton
- D. Atom
- It is the smallest constituent unit of ordinary matter that has the properties of a chemical element.
  - A. Neutron
  - Electron
  - C. Proton
  - D. Atom
- It is a subatomic particle, symbol p or p+, with a
  positive electric charge of +1e elementary
  charge and mass slightly less than that of a
  neutron.
  - A. Neutron
  - Electron
  - C. Protom
  - D. Atom
- It is a subatomic particle, symbol n or nO, with no net electric charge and a mass slightly larger than that of a proton.
  - A. Neutron
  - 8. Electron
  - C. Proton
  - D. Atom.
- 5. What type of nuclear reaction is this?



- A. Alpha Decay
- B. Beta Decay
- C. Gamma Ray
- D. No nuclear reaction

- It is a type of radioactive decay in which an atomic nucleus emits an alpha particle and thereby transforms or 'decays' into an atom with a mass number that is reduced by four and an atomic number that is reduced by two
  - A. Alpha Decay
  - B. Beta Decay
  - C. Gamma Ray
  - D. Radiation
- It is a type of radioactive decay in which a beta ray (fast energetic electron or positron) and a neutrino are emitted from an atomic nucleus.
  - A. Alpha Decay
  - B. Beta Decay
  - C. Gamma Ray
  - D. Radiation
- 8. What type of nuclear reaction is this?

$$30 \text{ Sr}$$
 →  $30 \text{ V}$  +  $9 \text{ e}$ 
 $230 \text{ Np}$  →  $230 \text{ Pu}$  +  $9 \text{ e}$ 
 $247 \text{ Cm}$  +  $9 \text{ e}$ 
 $367 \text{ F}$  →  $367 \text{ Cm}$  +  $9 \text{ e}$ 
 $367 \text{ F}$  →  $367 \text{ Cm}$  +  $9 \text{ e}$ 
 $367 \text{ Cm}$  +

- A. Alpha Decay
- B. Beta Decay
- C. Gamma Ray
- D. No nuclear reaction
- In atomic physics and quantum chemistry, the electron configuration is the distribution of electrons of an atom or molecule in atomic or molecular orbitals.
  - A. Electron Configuration
  - B. Periodic Table of Elements
  - C. Orbital Models
  - D. Emergy Levels
- 10. What type of nuclear reaction is this?

- A. Alpha Decay
- B. Beta Decay
- C. Gamma Ray
- D. No nuclear reaction

# **Physical Science 2june Exam Paper 1**

**LP Steffe** 

**Physical Science 2june Exam Paper 1:** 

#### Reviewing Physical Science 2june Exam Paper 1: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is truly astonishing. Within the pages of "**Physical Science 2june Exam Paper 1**," an enthralling opus penned by a highly acclaimed wordsmith, readers set about an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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