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| Clues | Vocabulary Terms |
|------------------------------------------------------------------------------------------------------------------------------------|------------------------|
| A type of bond that holds cations and anions together | SUDANNE CORNELIA I I I |
| A type of bond in which two atoms share a pair of tralence electrons | |
| A neutral group of atoms that are joined together by one or more covalent bonds | |
| A term describing a covalent bond in which electrons are not shared equally | |
| An ion that contains a covalently bonded group of atoms | |
| An ion with a negative charge | |
| An ion with a positive charge | |
| A notation that shows what elements a compound contains and the ratio of the atoms or ions of these elements in the compound | |
| Solids whose particles are arranged in a lattice structure | |
| A mixture of two or more elements, at least one of which is a metal. | |
| A type of bond that exists between a metal cation and the shared electrons that surround it | |
| | |

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