

Elements of the Sea

Ionic equations

- a. If 2 solutions react to form a solid, a precipitation reaction occurs. Eg. $\text{Ag}^+(\text{aq}) + \text{NO}_3^-(\text{aq}) + \text{Na}^+(\text{aq}) + \text{Br}^-(\text{aq})$ goes to $\text{AgBr}(\text{s}) + \text{Na}^+(\text{aq}) + \text{NO}_3^-(\text{aq})$

Then cancel spectators

$\text{Ag}^+(\text{aq}) + \text{Br}^-(\text{aq})$ goes to $\text{AgBr}(\text{s})$

- a. Always include state symbols
- a. Only split into ions if **aqueous and ionic**

Ionic solids

- a. Cations and anions are held together by strong electrostatic attractions in a **giant ionic lattice**.
- a. Each sodium ion is surrounded by 6 chloride ions **6:6 coordination**. It is a **simple cubic** structure.
- a. Ionic solids are hard and have a high melting point due to strong electrostatic attractions between ions.
- a. If ionic crystals contain water, the water molecules fit into the lattice the same way as regular ions. This is called the **water of crystallisation**. These crystals are said to be hydrated.
- a. If these hydrated crystals are heated, the water molecules are driven off and you are left with an **anhydrous** solid.



Ionic substances in solution

- a. As ionic substances dissolve in water, the ions become surrounded by water molecules and they spread throughout the solution. The hydrated ions are **randomly** arranged and behave **independently**.

Salters Elements Of Sea Module Test

Léo Pomar



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A. Lenihan,William Whigham Fletcher,1977 *The Chemical Engineer* ,1976 *Solar Energy Update* ,1978-04

Chemical Engineering Catalog ,1925 *SBASI* Stephen John Salter,1975 **Scientific and Technical Aerospace Reports** ,1979 *The Electrical Review* ,1906 SBASI S. J. Salter,Raymond Eugene Lundberg,G. L. McDougal,1975

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The Builder ,1881 *Energy Research Abstracts* ,1988 *The Builder* ,1881-07 **International Building Services Abstracts** ,1978 *International Conference on Future Energy Concepts, 30 January-1 February, 1979* Institution of Electrical Engineers. Science, Education, and Management Division,1979 Harper's Weekly ,1890 *Machinery Lloyd* ,1953

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In an electronic digital era where connections and knowledge reign supreme, the enchanting power of language has been apparent than ever. Its power to stir emotions, provoke thought, and instigate transformation is truly remarkable. This extraordinary book, aptly titled "**Salters Elements Of Sea Module Test**," written by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we shall delve to the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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