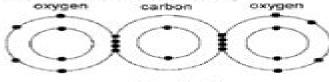


MERCY JUNIOR COLLEGE Grade 11 Physical Science Finals Exam

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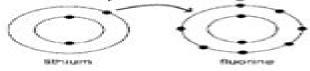
- I. MULTIPLE CHOICE INSTRUCTION: Encircle the letter of the correct answer. (1 POINT EACH)
- 1. It is a subatomic particle, symbol e- or
 - β-, with a negative elementary electric charge.
 - A. Neutron
 - 8. Electron
 - C. Protom
 - D. Attorni
- It is a subatomic particle, symbol p or p+, with a
 positive electric charge of +1e elementary
 charge and mass slightly less than that of a
 meutron.
 - A. Neutron
 - 8. Electron
 - C. Protom
 - Atom
- It is a subatomic particle, symbol n or nO, with no net electric charge and a mass slightly larger than that of a proton.
 - A. Neutron B. Electron
 - C. Proton
 - D. Atom
- 4. Is this an example of ionic bonding?



carbon dioxide

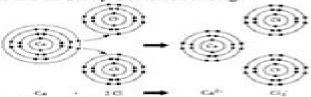
- A. Yes, it is ionic bonding.
- No, it is of another type of chemical bonding.
- C. There is no chemical bonding present.
- D. This is just an orbital model.

5. Is this an example of ionic bonding?



littreum fluoride

- A. Yes, it is ionic bonding.
- No, it is of another type of chemical bonding.
- There is no chemical bonding present.
- D. This is just an orbital model.
- In atomic physics and quantum chemistry, the electron configuration is the distribution of electrons of an atom or molecule in atomic or molecular orbitals.
 - A. Electron Configuration
 - Periodic Table of Elements
 - C. Orbital Models
 - D. Emergy Levels
- 7. Is this an example of ionic bonding?



- A. Yes, it is ionic bonding.
- No, it is of another type of chemical bonding.
- There is no chemical bonding present.
- This is just an orbital model.
- It is a chemical rule of thumb that reflects observation that atoms of main-group elements tend to combine in such a way that each atom has eight electrons in its valence shell, giving it the same electronic configuration as a noble gas.
 - A. Octet Bulle
 - 8. Valence Electrons
 - C. Atomic Orbitals
 - D. Chemical Bonds
- It is the smallest constituent unit of ordinary matter that has the properties of a chemical element.
 - A. Nieutron
 - B. Electron
 - C. Proton
 - D. Atom
- 10. It is a type of chemical bond that involves the electrostatic attraction between oppositely

Physical Science Grade 11 Final Examination 2014 Scope

Michael Brown

Physical Science Grade 11 Final Examination 2014 Scope:

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Table of Contents Physical Science Grade 11 Final Examination 2014 Scope

- 1. Understanding the eBook Physical Science Grade 11 Final Examination 2014 Scope
 - The Rise of Digital Reading Physical Science Grade 11 Final Examination 2014 Scope
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Physical Science Grade 11 Final Examination 2014 Scope
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Physical Science Grade 11 Final Examination 2014 Scope
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Physical Science Grade 11 Final Examination 2014 Scope
 - Personalized Recommendations

- Physical Science Grade 11 Final Examination 2014 Scope User Reviews and Ratings
- Physical Science Grade 11 Final Examination 2014 Scope and Bestseller Lists
- 5. Accessing Physical Science Grade 11 Final Examination 2014 Scope Free and Paid eBooks
 - Physical Science Grade 11 Final Examination 2014 Scope Public Domain eBooks
 - Physical Science Grade 11 Final Examination 2014 Scope eBook Subscription Services
 - Physical Science Grade 11 Final Examination 2014 Scope Budget-Friendly Options
- 6. Navigating Physical Science Grade 11 Final Examination 2014 Scope eBook Formats
 - o ePub, PDF, MOBI, and More
 - Physical Science Grade 11 Final Examination 2014 Scope Compatibility with Devices
 - Physical Science Grade 11 Final Examination 2014 Scope Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Physical Science Grade 11 Final Examination 2014 Scope
 - Highlighting and Note-Taking Physical Science Grade 11 Final Examination 2014 Scope
 - Interactive Elements Physical Science Grade 11 Final Examination 2014 Scope
- 8. Staying Engaged with Physical Science Grade 11 Final Examination 2014 Scope
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Physical Science Grade 11 Final Examination 2014 Scope
- 9. Balancing eBooks and Physical Books Physical Science Grade 11 Final Examination 2014 Scope
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Physical Science Grade 11 Final Examination 2014 Scope
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Physical Science Grade 11 Final Examination 2014 Scope
 - Setting Reading Goals Physical Science Grade 11 Final Examination 2014 Scope
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Physical Science Grade 11 Final Examination 2014 Scope
 - Fact-Checking eBook Content of Physical Science Grade 11 Final Examination 2014 Scope

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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