

A chemical reaction is a transformation of a chemical reaction in which the number of atoms of each element is conserved.

A coefficient represents the number of atoms of a particular element in a chemical reaction. A coefficient is placed directly in front of a chemical formula.

A subscript is placed after the element symbol to indicate the number of atoms of that element in a chemical formula.

Why is it important to balance chemical equations?

Chemical reactions are balanced to show that the number of atoms of each element is the same on both sides of the equation.

What does the total say is important to remember when balancing equations?

Remember, the total number of atoms of each element must be the same on both sides of the equation.

### Week 8, Day 2

The number of protons in an atom cannot change (it is fixed) but a different number of neutrons can change.

The most stable atomic configuration has a nucleus with a balanced number of protons and neutrons (equal numbers of protons and neutrons).

How does a positive ion form from a neutral atom? What is another name for positive ion?

When an atom loses one or more electrons, it has more protons than it does electrons, so it is no longer neutral. This atom has a positive charge and is called a positive ion.

How does a negative ion form from a neutral atom? What is another name for a negative ion?

When an atom gains one or more electrons, it has more electrons than it does protons, so it is no longer neutral. This atom has a negative charge and is called a negative ion.



Atoms in Group 1A have 1 electron in their outer shell and Group 2A atoms have 2 electrons in their outer shell. To become more stable (like the noble gases), the outermost electrons are easily given up (they become positively charged). For example, Group 1A atoms in their neutral state and easily lose 1 electron to form the most stable electron configuration of the noble gas next to them.



Atoms are formed by taking the first part of the element name and adding the suffix -ide. For example, sodium chloride, NaCl, is called a sodium ion.



When sodium and chlorine atoms react, the sodium atom gives up the outermost electron, becoming a cation (positively charged ion), and the chlorine atom accepts that electron, becoming an anion (negatively charged ion).

The positively charged sodium ion and the negatively charged chloride ion are attracted to each other. The bond formed between the positively charged ion, holding the two ions together.

# Physical Science Exploring Space Notetaking Guide Answer

**K Morrison**



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