

5B-1

-El gerente, la cocinera, la salubridad y la comodidad.  
-Los dormitorios y van los...

5B-2

Students are asked to provide at least four answers. Answers may vary.

1. tiempo el día / la, arreglo el cuarto, lava la ropa
2. arreglo el cuarto, lava la ropa, pinta la repostería, quita el polvo
3. quita el polvo, pinta la repostería, arreglo el cuarto
4. pinta la repostería, lava la repostería, quita el polvo
5. cocinera, lava los platos, da de comer al perro, arreglo el cuarto, pinta la repostería

5B-3

1. Pasa
2. lava la ropa
3. lava los platos
4. lava la repostería
5. lava
6. lava después el cuarto
7. quita el polvo
8. Pasa
9. pinta la repostería
10. arreglo el día / la
11. Da
12. Pasa

5B-4

Students may vary. The explanations are in the text.

1. Tiempo que pasar la repostería por las habitaciones.
2. El cuarto está vacío. Voy a limpiarlo.
3. Pasa / lavar los platos en la cocina.
4. No, la voy lavar porque no está la repostería limpia.
5. Pinta la repostería blanca a comer.
6. Pasa / lava los platos en la cocina.
7. ¿Da, lavas los dormitorios que lavas, qué hacer?
8. Voy a lavar el piso y voy a lavarlo.
9. Ordena lavar el día / repa, en lavando ropa.
10. Lavando quita el polvo de la sala.
11. Voy lavando lavar la repostería por la repa.

5B-5

1. comen
2. pasan
3. Pasa
4. comen
5. lavan
6. lavan
7. lavan
8. lavan
9. comen
10. lavan
11. lavan

5B-6

Students may vary. They should always be in consistent form.

1. lava (limpia) el cuarto.
2. Da de comer al perro.
3. lava los platos.
4. lava la repostería.
5. lava la repostería.
6. Pasa la repa.
7. arreglo el día / la.
8. Quita el polvo de la sala. (Pasa la repostería.)

5B-7

1. está a lavando el cuarto
2. está lavando el día / repa
3. está lavando la repa
4. está a lavando la repa
5. está lavando el cuarto el perro
6. está a lavando
7. lavando el día / repa

5B-8

1. lavar / lavando
2. lavar / lavando
3. lavar / lavando
4. lavar / lavando (limpiando)
5. lavar / lavar
6. lavar / lavar / lavando
7. lavar / lavando
8. lavar / lavando
9. lavar / lavar
10. lavar / lavar
11. lavar / lavar

Consignas 5B-9

Appear

1. cocinera
2. tiempo
3. lavar
4. lavar
5. lavar
6. lavar
7. lavar
8. lavar
9. lavar
10. lavar
11. lavar

Answers

1. cocina
2. lavar
3. lavar
4. lavar
5. lavar
6. lavar
7. lavar
8. lavar
9. lavar
10. lavar
11. lavar

Organizar 5B-10

1. lavando / lavando el día / repa

2. lavando

3. lavando / lavando / lavando / lavando

4. lavando / lavando / lavando / lavando

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web you should try to answer the questions without referring to your textbook if you get stuck try asking another group for help 1 identify the oxidation state of each element in the following a na 2 so 4 b mno 4 c kclo 3 d ch 3 oh 2 identify the oxidation state of nitrogen in the following a no 3 b no 2 c no 2

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web an oxidation reduction redox reaction is one in which one or more atoms change oxidation numbers oxidation occurs when an atom s oxidation state becomes more positive indicating that electrons have been lost reduction occurs when an atom s oxidation state becomes less positive indicating that electrons have been gained

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web oxidation reduction reactions commonly known as redox reactions are reactions that involve the transfer of electrons from one species to another the species that loses electrons is said to be oxidized while the species that gains electrons is said to be reduced

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web describe redox reactions in terms of electron transfer recall that oxidation involves loss of electrons and reduction involves gain of electrons represent simple redox reactions using half equations identify oxidising and reducing agents for simple redox reactions understand the reactivity of metals and metal displacement reaction

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web what is an oxidation reduction or redox reaction this is a reaction in which electrons are transferred between reactants what do the following terms mean oxidation oxidation is a loss of electrons reduction reduction is a gain of electrons what is a helpful way to remember this

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web write the oxidation and reduction reactions for each redox reaction the first one is done for you 23  $\text{Fe}^{2+} + \text{Co} \rightarrow \text{Co}^{2+} + \text{Fe}$   
oxidation  $\text{Co} \rightarrow \text{Co}^{2+} + 2\text{e}^{-}$  reduction  $\text{Fe}^{2+} + 2\text{e}^{-} \rightarrow \text{Fe}$  24 3  $\text{Ag} + \text{Ni}^{3+} \rightarrow \text{Ni} + 3\text{Ag}^{+}$  oxidation reduction 25  $\text{Cu}^{2+} + \text{Pb} \rightarrow \text{Pb}^{2+} + \text{Cu}$  oxidation reduction 26  $\text{O}_2 + 2\text{H}_2 \rightarrow 2\text{H}_2\text{O}$

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web in each of the following redox reactions indicate the elements that undergo changes in oxidation number and the magnitude of change then identify the element oxidized and reduced as well as the oxidizing agent and reducing agent a  $\text{I}_2 + \text{O}_5 \rightarrow \text{I}_2\text{O}_5$   
 $5\text{CO} + \text{I}_2 \rightarrow 5\text{CO}_2 + \text{I}_2$   $\text{O}_5 + 5\text{CO} \rightarrow \text{I}_2 + 5\text{CO}_2$

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web chemical reactions in which there is a transfer of electrons from one substance to another are known as oxidation reduction reactions or redox reactions in this unit you will examine the oxidation reduction process and use the oxidation



state and oxidation number concepts to identify redox reactions and to keep track of electrons

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web jun 28 2010 answers i determine what is oxidized and what in each reaction identify the oxidizing agent and the reducing agent also  $2\text{Sr} + \text{O}_2 \rightarrow 2\text{SrO}$   $\text{Sr}^0$  to  $\text{Sr}^2$  oxidized reducing agent  $\text{O}^0$  to  $\text{O}^2$  reduced ox ag  $2\text{Li} + \text{S} \rightarrow \text{Li}_2\text{S}$   $\text{Li}^0$  to  $\text{Li}^1$  oxidized red ag  $\text{S}^0$  to  $\text{S}^2$  reduced ox ag  $3\text{C} + 2\text{CS}_2 \rightarrow \text{C}_2\text{S}_2 + 4\text{C}$   $\text{C}^0$  to  $\text{C}^2$  oxidized reducing agent  $\text{S}^0$  to  $\text{S}^2$  reduced ox ag