# Common Polyatomic Ions

acetate	C2H3O2
ammonium	NH4*
arsenate	AsO <sub>4</sub> 3-
arsenite	AsO <sub>3</sub> 3-
azide	N <sub>3</sub> <sup>-</sup>
benzoate	C7H5O2
borate	BO33-
bromate	BrO <sub>3</sub>
carbonate	CO32-
chlorate	CIO <sub>3</sub>
chlorite	CIO <sub>2</sub>
chromate	CrO <sub>4</sub> <sup>2-</sup>
cyanide	CN-
dichromate	Cr2O72-
dihydrogen phosphate	H <sub>2</sub> PO <sub>4</sub>
dihydrogen phosphite	H <sub>2</sub> PO <sub>3</sub> <sup>-</sup>
hydrogen carbonate	HCO <sub>3</sub> T
hydrogen phosphate	HPO42-
hydrogen phosphite	HPO32-
hydrogen sulfate	HSO <sub>4</sub>
hydrogen sulfide	HS-
hydrogen sulfite	HSO <sub>3</sub> -
hydroxide	OH-
hypochlorite	CIO-
iodate	1O <sub>3</sub> -
manganate	MnO <sub>4</sub> 2-
nitrate	NO <sub>3</sub>
nitrite	NO <sub>2</sub> T
oxalate	C2O42-
perchlorate	CIO <sub>4</sub> -
permanganate	MnO <sub>4</sub>
peroxide	O22-
phosphate	PO43-
phosphite	PO33-
silicate	SiO <sub>3</sub> <sup>2-</sup>
sulfate	SO42-
sulfite	SO32-
tartrate	C4H4O62-
thiocyanate	SCN-
thiosulfate	S2O32-

	9
AsO <sub>3</sub> <sup>3</sup> -	arsenite
AsO <sub>4</sub> 3-	arsenate
BO <sub>3</sub> 3-	borate
BrO <sub>3</sub>	bromate
C2H3O2	acetate
C2O42-	oxalate
C4H4O62-	tartrate
C7H5O2	benzoate
CIO-	hypochlorite
CIO <sub>2</sub> <sup>-</sup>	chlorite
CIO <sub>3</sub> <sup>-</sup>	chlorate
CIO <sub>4</sub> -	perchlorate
CN-	cyanide
CO32-	carbonate
Cr <sub>2</sub> O <sub>7</sub> <sup>2-</sup>	dichromate
CrO <sub>4</sub> <sup>2-</sup>	chromate
H <sub>2</sub> PO <sub>3</sub> <sup>-</sup>	dihydrogen phosphite
H <sub>2</sub> PO <sub>4</sub> <sup>-</sup>	dihydrogen phosphate
HCO <sub>3</sub>	hydrogen carbonate
HPO <sub>3</sub> <sup>2-</sup>	hydrogen phosphite
HPO <sub>4</sub> <sup>2-</sup>	hydrogen phosphate
HS-	hydrogen sulfide
HSO <sub>3</sub>	hydrogen sulfite
HSO <sub>4</sub>	hydrogen sulfate
IO <sub>3</sub>	iodate
MnO <sub>4</sub>	permanganate
MnO <sub>4</sub> <sup>2-</sup>	manganate
N <sub>3</sub> <sup>-</sup>	azide
NH4*	ammonium
NO <sub>2</sub> T	nitrite
NO <sub>3</sub>	nitrate
O22-	peroxide
OH-	hydroxide
PO33-	phosphite
PO <sub>4</sub> 3-	phosphate
S2O32-	thiosulfate
SCN	thiocyanate
SiO <sub>3</sub> <sup>2-</sup>	silicate
SO32-	sulfite
SO <sub>4</sub> <sup>2-</sup>	sulfate

# **Physical Science Formulas With Polyatomic Ions Answer**

**Sebastian Brünink** 

**Physical Science Formulas With Polyatomic Ions Answer:** 

#### Physical Science Formulas With Polyatomic Ions Answer Book Review: Unveiling the Magic of Language

In an electronic digital era where connections and knowledge reign supreme, the enchanting power of language has be more apparent than ever. Its ability to stir emotions, provoke thought, and instigate transformation is truly remarkable. This extraordinary book, aptly titled "**Physical Science Formulas With Polyatomic Ions Answer**," compiled by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we shall delve into the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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