

Answer all questions on the paper/answer book provided.
A scientific calculator may be used.

QUESTION 1

There are four possible options for each answer in the following questions. Each question has only ONE correct answer. Choose the correct answer and write only A, B, C or D next to the question number.

- 1.1 A compound consists of ions X^{2+} and Y^{3-} . A possible formula for the compound could be:
- A $3X2Y$
 - B X_2Y_3
 - C X_3Y_2
 - D $(XY)_6$
- (2)
- 1.2 Which of the following is not a characteristic of a physical change:
- A no new products are formed
 - B changes can be reversed
 - C observable with the naked eye
 - D the parts of the substance lose their chemical identity
- (2)
- 1.3 Ionic bonding occurs between:
- A two non-metals
 - B two metals
 - C a metal and a non-metal
 - D a metal and a semi-metal
- (2)
- 1.4 Khumalo heats chocolate in a double boiler and observes the changes. The most accurate conclusion to his experiment is:
- A Chocolate undergoes a chemical change when it is heated
 - B Chocolate undergoes a physical change when it is heated
 - C Chocolate decomposes into its constituent components when heated
 - D Chocolate gives off latent energy when it melts
- (2)

Physical Science Control Test Term 3

Marcel A. Müller



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Table of Contents Physical Science Control Test Term 3

1. Understanding the eBook Physical Science Control Test Term 3
 - The Rise of Digital Reading Physical Science Control Test Term 3
 - Advantages of eBooks Over Traditional Books
2. Identifying Physical Science Control Test Term 3
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Physical Science Control Test Term 3
 - User-Friendly Interface

4. Exploring eBook Recommendations from Physical Science Control Test Term 3
 - Personalized Recommendations
 - Physical Science Control Test Term 3 User Reviews and Ratings
 - Physical Science Control Test Term 3 and Bestseller Lists
5. Accessing Physical Science Control Test Term 3 Free and Paid eBooks
 - Physical Science Control Test Term 3 Public Domain eBooks
 - Physical Science Control Test Term 3 eBook Subscription Services
 - Physical Science Control Test Term 3 Budget-Friendly Options
6. Navigating Physical Science Control Test Term 3 eBook Formats
 - ePub, PDF, MOBI, and More
 - Physical Science Control Test Term 3 Compatibility with Devices
 - Physical Science Control Test Term 3 Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Physical Science Control Test Term 3
 - Highlighting and Note-Taking Physical Science Control Test Term 3
 - Interactive Elements Physical Science Control Test Term 3
8. Staying Engaged with Physical Science Control Test Term 3
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Physical Science Control Test Term 3
9. Balancing eBooks and Physical Books Physical Science Control Test Term 3
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Physical Science Control Test Term 3
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Physical Science Control Test Term 3
 - Setting Reading Goals Physical Science Control Test Term 3
 - Carving Out Dedicated Reading Time

12. Sourcing Reliable Information of Physical Science Control Test Term 3
 - Fact-Checking eBook Content of Physical Science Control Test Term 3
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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