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Matr.:

Klausur: Physikalische Chemie

Physikalische Chemie: Chapter 10: Outer Thermal Energy and Mass

Multiple Choice:

1. The Boltzmann constant is defined as:
- a) $k_B = 1.38 \times 10^{-23} \text{ J/K}$
 - b) $1.38 \times 10^{-23} \text{ J/K}$
 - c) $1.38 \times 10^{-23} \text{ J/K}$
 - d) $1.38 \times 10^{-23} \text{ J/K}$
2. Energy added to a system:
- a) increases the system's internal energy
 - b) decreases the system's internal energy
 - c) increases the system's temperature
 - d) decreases the system's temperature
3. When a system is heated, the temperature of the system:
- a) increases
 - b) decreases
 - c) remains constant
 - d) increases or decreases depending on the system
4. The heat capacity of a system is defined as:
- a) the amount of heat added to the system
 - b) the amount of heat removed from the system
 - c) the amount of heat added to the system divided by the change in temperature
 - d) the amount of heat removed from the system divided by the change in temperature
5. The heat capacity of a system is defined as:
- a) the amount of heat added to the system
 - b) the amount of heat removed from the system
 - c) the amount of heat added to the system divided by the change in temperature
 - d) the amount of heat removed from the system divided by the change in temperature
6. The heat capacity of a system is defined as:
- a) the amount of heat added to the system
 - b) the amount of heat removed from the system
 - c) the amount of heat added to the system divided by the change in temperature
 - d) the amount of heat removed from the system divided by the change in temperature
7. The heat capacity of a system is defined as:
- a) the amount of heat added to the system
 - b) the amount of heat removed from the system
 - c) the amount of heat added to the system divided by the change in temperature
 - d) the amount of heat removed from the system divided by the change in temperature
8. The heat capacity of a system is defined as:
- a) the amount of heat added to the system
 - b) the amount of heat removed from the system
 - c) the amount of heat added to the system divided by the change in temperature
 - d) the amount of heat removed from the system divided by the change in temperature

Physical Science Chapter 16 Test

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