FUNCTIONAL GROUPS OF ORGANIC COMPOUNDS

-	Homologous Structure of functional		Example c	of a compound
	Homologous series	group	Name	Structural formula
	alkanes	Only C-H and C-C single bonds	ethane .	н
	alkenes	`c=c(ethene)c=<,
	alkynes	-c≡c-	ethyne	н-с=с-н
	haloalkanes (alkyl halides)		bromoethane	H—————————————————————————————————————
	alcohols (alkanols)	-ţ-о-н	ethanol	H
	aldehydes	н	ethanal	н—Ё—н
	ketones	-!-9-!-	proper-2-one	
Ī	carboxylic acids	_2-о-н	ethanoic acid	н-6-8-0-н
Ī	esters		methyl ethancate	H

NAMING OF ORGANIC COMPOUNDS

The name of every organic molecule has three parts:

- The parent name indicates the number of C atoms in the longest carbon chain in the molecule.
- The suffix indicates what functional group is present.
- The prefix reveals the identity, location and number of substituents attached to the carbon chain.

What and where are the substituents?

parent How many carbons?

Suffix
What is the functional group/homologous series?

di - 2 tri - 3 telra - 4

Number of carbon atoms	Parent name	Number of carbon atoms	Parent name
1	meth	5	pent
2	eth	6	hex
3	prop	7	hept
4	but	8	oct

prefix, suffix and parent!

Page 4 of 41 Tetra – Chloro – melta

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Cybellium

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