

PHYSICS I: Fluid Mechanics

Class notes: Topic, rules, and formula. From the formula given, you can derive the equations.

- ❖ The **density** of a substance is simply the ratio of its mass to its volume. It also dictates if the substance (solid or fluid) sinks or floats in a mixture with a fluid.

$$\rho = \frac{m}{V}$$

- ❖ The **specific density** (also referred to as *relative density* or *specific gravity*) is a dimensionless quantity that relates the density of a substance relative to a reference.

$$\text{specific gravity} = \frac{\rho(\text{substance})}{\rho(\text{reference})}$$

- ❖ For solids and liquids, the reference substance is water at 4 °C ($\rho = 1000 \text{ kg/m}^3$). For gases, the reference substance used is oxygen at 0 °C and 1 atm pressure ($\rho = 1.43 \text{ kg/m}^3$).

- ❖ **Pressure** refers to the force exerted per unit area. For fluids, this physical quantity is dependent on the fluid height, not volume.

$$\text{Pressure} = \frac{\text{Force}}{\text{Area}}$$

- ❖ For fluids, pressure values vary depending on the method used. Typical pressure readings report the fluid's **gauge pressure**, whereas the fluid's **absolute pressure** is the sum of the gauge pressure and the pressure of its environment (typically air).

$$\text{Gauge pressure} \quad P_g = \rho gh$$

$$\text{Absolute pressure} \quad P_{\text{abs}} = P_g + \rho gh$$

$$\text{Barometric pressure}$$

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Table of Contents Physics Fluid Mechanics Study Guide

1. Understanding the eBook Physics Fluid Mechanics Study Guide
 - The Rise of Digital Reading Physics Fluid Mechanics Study Guide
 - Advantages of eBooks Over Traditional Books
2. Identifying Physics Fluid Mechanics Study Guide
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Physics Fluid Mechanics Study Guide
 - User-Friendly Interface
4. Exploring eBook Recommendations from Physics Fluid Mechanics Study Guide
 - Personalized Recommendations

- Physics Fluid Mechanics Study Guide User Reviews and Ratings
- Physics Fluid Mechanics Study Guide and Bestseller Lists
- 5. Accessing Physics Fluid Mechanics Study Guide Free and Paid eBooks
 - Physics Fluid Mechanics Study Guide Public Domain eBooks
 - Physics Fluid Mechanics Study Guide eBook Subscription Services
 - Physics Fluid Mechanics Study Guide Budget-Friendly Options
- 6. Navigating Physics Fluid Mechanics Study Guide eBook Formats
 - ePub, PDF, MOBI, and More
 - Physics Fluid Mechanics Study Guide Compatibility with Devices
 - Physics Fluid Mechanics Study Guide Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Physics Fluid Mechanics Study Guide
 - Highlighting and Note-Taking Physics Fluid Mechanics Study Guide
 - Interactive Elements Physics Fluid Mechanics Study Guide
- 8. Staying Engaged with Physics Fluid Mechanics Study Guide
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Physics Fluid Mechanics Study Guide
- 9. Balancing eBooks and Physical Books Physics Fluid Mechanics Study Guide
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Physics Fluid Mechanics Study Guide
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Physics Fluid Mechanics Study Guide
 - Setting Reading Goals Physics Fluid Mechanics Study Guide
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Physics Fluid Mechanics Study Guide
 - Fact-Checking eBook Content of Physics Fluid Mechanics Study Guide

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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