

where k is the electrostatic force constant. The value of k depends on the nature of the medium between the two charges and the system of units chosen to measure F , q_1 , q_2 and r .

For two charges located in free space and in SI units, we have,

$$k = \frac{1}{4\pi\epsilon_0} = 9 \times 10^9 \text{ Nm}^2\text{C}^{-2} \quad (2)$$

where ϵ_0 is the permittivity of free space.

- **Unit of charge:**

1. In SI system: The SI unit of charge is coulomb. In the above equation, if $q_1 = q_2 = 1\text{C}$ and $r = 1\text{m}$, then

$$F = \frac{1}{4\pi\epsilon_0} = 9 \times 10^9 \text{ N} \quad (3)$$

Therefore, one coulomb is that amount of charge that repels an equal and similar charge with a force of $9 \times 10^9 \text{ N}$ when placed in vacuum at a distance of one metre from it.

2. In CGS system: The CGS unit of charge is statcoulomb.
1 coulomb = 3×10^9 statcoulomb = 3×10^9 e.s.u. of charge. [e.s.u. = electrostatic unit]
one e.s.u. or one statcoulomb is that charge which repels an identical charge in vacuum at a distance of 1 centimetre from it with a force of one dyne.
3. Electromagnetic unit (e.m.u.): e.m.u. is the electromagnetic unit of charge or abcoulomb.
1 coulomb = $\frac{1}{10}$ abcoulomb = $\frac{1}{10}$ e.m.u. of charge.

- **Coulomb's law in vector form:** In vector form coulomb's law may be expresses as,

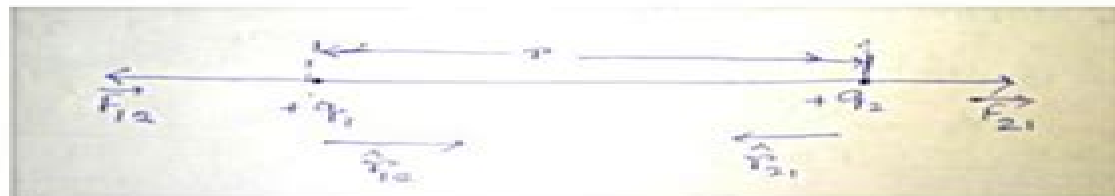


Figure 2: Two stationary point charge separated by a distance r (vector form).

$\vec{F}_{21}$ = force on charge q_2 due to charge q_1

$$\vec{F}_{21} = \frac{1}{4\pi\epsilon_0} \frac{q_1 q_2}{r^2} \hat{r}_{12} \quad (4)$$

where, $\hat{r}_{12} = \frac{\vec{r}_{12}}{r}$, is a unit vector in the direction from q_1 to q_2 .

The coulombian force between unlike charges ($q_1 q_2 < 0$) are attractive.

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