



NATIONAL
SENIOR CERTIFICATE

GRADE 11

NOVEMBER 2020

PHYSICAL SCIENCES P2
(CHEMISTRY)
EXEMPLAR

Duration: 120
Time: 1 hour



The question paper consists of 12 pages, including 4 extra sheets.
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QUESTION 8

PHYSICAL SCIENCES PAPER 2

ACID AND BASES

GRADE 11

NOVEMBER 2020

QUESTION 8: ACID AND BASES (10)

8.1. Draw a potential energy diagram for the reaction represented by

The following equation shown on the page:

a. Heat of this reaction (20)
b. Activation energy (10)

QUESTION 8 (Start on a NEW page)

8.2. Sodium hydroxide is an ionic compound with the following percentage composition:

Na 40.1%, O 52.0%, H 7.9%

The molar mass of the compound is 40 g mol⁻¹

8.2.1. Calculate the mass percentage of Na (10)
8.2.2. Calculate the number of moles of NaOH in 40 g (10)
8.2.3. Calculate the volume of NaOH in 40 g (10)

8.3. Concentrated sulphuric acid is a mixture of sulphuric acid and water. It is a mixture of 98% sulphuric acid and 2% water. It is a mixture of 98% sulphuric acid and 2% water.

8.3.1. Calculate the concentration of the sulphuric acid in mol dm⁻³ (10)
8.3.2. Calculate the concentration of the sulphuric acid in g dm⁻³ (10)

8.4. Sodium hydroxide is used in the manufacture of soap. The following reaction takes place:

$\text{C}_{18}\text{H}_{37}\text{O}_2 + \text{NaOH} \rightarrow \text{C}_{18}\text{H}_{35}\text{O}_2 + \text{H}_2\text{O}$

Calculate the volume of hydrogen gas (H₂) that would be produced at STP if 10 g of sodium hydroxide reacts completely. (10)



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