

- 3.6 What type of chemical bond is present in NH_3 ? Draw Lewis electron dot diagram of NH_3 .
(The atomic number of H and N are 1 and 7 respectively) 2
- 3.7 How is ionic bond formed in sodium fluoride?
(The atomic number of F and Na are 9 and 11 respectively) 2
- OR**
- C has 4 electrons in its outermost shell and O has 6 electrons in its outermost shell. Draw the Lewis dot diagram of CO_2 molecule. 2
- 3.8 Distinguish between naphthalene and sodium chloride by two physical properties <https://www.westbengalboard.com> 2
- 3.9 What is meant by strong electrolyte? 2
- OR**
- What are meant by cathode and anode electrodes? 2

Group 'D'

4. Answer the following questions (alternatives are to be noted) :
- 4.1 Establish the combined form of Boyle's law and Charles' law. 3
- 4.2 By the reaction of aluminium with 40 grams of metallic oxide and high temperature, 28 grams of the metal and 25.5 grams of aluminium oxide are produced. How many grams of aluminium are required for the reaction? 3
- OR**
- The chemical equation for the reaction between sulphuric acid and sodium hydroxide is:
- $$\text{H}_2\text{SO}_4 + 2\text{NaOH} \longrightarrow \text{Na}_2\text{SO}_4 + 2\text{H}_2\text{O}$$
- How many grams of sodium hydroxide will be required to react completely with 4.9 grams of sulphuric acid?
(H = 1, O = 16, Na = 23, S = 32) 3
- 4.3 A light is incident from air medium on another medium. If the refractive index of this medium is 1.5 and the wavelength of light in this medium is $4000\text{\AA}$, what is the wavelength of light in the air medium? What is the velocity of the light in this medium? 3

OR

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