(2)

(1)

(3) [13]

QUESTION 2

2.1 The graph of molecular size versus the boiling point is given below.

The letters A, B and C represent the compounds CH₄, C₄H₆ and C₃H₈ respectively.



- 2.1.1 Define the term boiling point.
- 2.1.2. Describe the trend in the boiling points of the compounds as shown by the graph.
- 2.1.3. Explain the answer to QUESTION 2.1.2 by referring to MOLECULAR SIZE, TYPE and STRENGTH of INTERMOLECULAR FORCES. (3)
- 2.1.4 Which ONE of the compounds (A, B or C) has the HIGHEST vapour pressure? Explain the answer by referring to the data on the graph.
 (2)
- 2.2 Consider the two molecules in the table below.

NAME OFSUBSTANCE	FORMULA	MOLACULAR MASS (g)	BOILING POINT
Ammonia	NH ₃	17	- 33
Phosphine	PH ₃	×	- 87.4

- 2.2.1 Determine the molecular mass (x) of phosphine
- 2.2.2 Explain the difference in the boiling points by referring to the TYPE and STRENGTH of INTERMOLECULAR FORCES.

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guideline for section 12 of ontario regulation 22 04 electrical - Mar 19 2022

web 1 2 18 oesc means the ontario electrical safety code incorporated by reference under ontario regulation 164 99 1 2 19 ohsa means the occupational health and safety act r s o 1990 c o 1 1 2 20 primary distribution line

2021 ontario electrical safety code 28th edition esa - Mar 31 2023

web mississauga on november 12 2021 the electrical safety authority esa would like to advise that the minister of government and consumer services has approved the regulatory amendment to adopt the 2021 ontario electrical safety code oesc which comes into effect on may 5 2022