

Reflections on my Personal and Professional Growth

9 pages | 2163 words

My time studying with the University of Phoenix online has impacted my life in many ways. I have had the possibility to find out a lot by using the Internet to help me further my education. I feel that my degree will open numerous brand-new doors for me, however first I will return and discuss my impressions from when I first started taking nursing classes online. Learning has always been extremely crucial to me. I understand that we do not get really far in this world if we are not happy to take instruction and utilize it to the very best of our capabilities. Being a student was an amazing chance, specifically in the major field that I had actually picked, nursing. Nursing is an ever-changing field, and it is extremely essential to keep up with the latest information if one is going to be expert and the best they can be in their selected profession. I feel that strengthening my understanding base made me grow in my individual life due to the fact that there was a lot that I might achieve simply by taking the actions to find out more. In nursing learning more leads to larger and higher opportunities and the chance to take jobs that might be out of reach if I did pass by to acquire higher knowledge. While discovering new things might be difficult sometimes, I feel that the understanding gotten is all worth it in the end. I first discovered the English language nine years back when I immigrated to the United States. My issue solving skills and my oral communication abilities were terrific, however I had some trouble with my writing abilities. As for recovering details and using it, I had no issue finding information, whether in books or on the Web, and utilizing it for any job that was put in front of me. Partnership was not an issue as long as it was performed in an oral manner. As I have actually mentioned, my writing abilities were only reasonable, therefore collaboration involving writing could be fairly hard. My career goals were based around degrees that I had actually already made, specifically a registered nurse degree and a breathing therapist degree. My immediate objective was to get my Bachelors of Science degree in nursing, however, I also had other goals. For example, I wished to take my 2 degrees that were already earned and utilize them to enter a nurse anesthesia school. My most lofty objective was to finish my bachelor's degree and begin taking more classes to earn a Masters degree one day. As you may see, I am very dedicated to my chosen line of work and desire to be the best at it. Education is my ticket to being the best, and I crave all kinds of new information and knowledge that will make me an excellent nurse. While I was in the program, I found that the classes raised my level of competence in several areas. While I was already quite good at problem solving and oral communication, taking classes that challenged me and required me to...

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for the nitration 4.01 ml of nitric acid 4.00 ml of sulfuric acid and 4.56 ml bromobenzene were used to complete the reaction with these measurements bromobenzene was found to be the limiting reagent since only 4.56 ml of bromobenzene was used the amount of product predicted to be recovered was 8.7533 g

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the aim of the experiment is to determine and study electrophilic aromatic substitution reaction especially nitration of methyl benzoate to give methyl m-nitrobenzoate thus determine the weight melting point and percentage yield of the pure methyl m-nitrobenzoate abstract

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in this experiment we will carry out the nitration of bromobenzene bromine substituted on an aromatic ring is an ortho/para directing group it will direct an incoming electrophile to the ortho and para positions it is also deactivating meaning that its reactions will be slower than the identical reactions of benzene

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section 20711 lab experiment nitration of bromobenzene introduction the chemical reaction of nitration consists of a nitro group being added to or substituted in a molecule nitration can basically be carried out by a mixture of concentrated nitric acid and sulfuric acid this mixture is useful to obtain the active nitronium ion

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1 go to the restroom and wash the affected area with lots of soap and water the soap in the lab is not suitable for this purpose 2 return to the lab and apply mineral oil to the affected area 3 the summary to this warning is not to touch any part of your body while performing this experiment if you must leave the lab for any reason 1

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in the nitration of bromobenzene a mixture of sulfuric acid and nitric acid react with one another nitric acid the weaker of the two acids will serve as base to form nitronium ion bromobenzene will attack the nitronium ion by nucleophilic attack

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nitration of bromobenzene lab report improved essays 731 words 3 pages open document essay sample check writing quality show more this experiment was performed in order to demonstrate the principles of nitration through electrophilic aromatic substitution it also examined the effects of bromine on an aromatic ring ortho/para vs meso directing

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then actual is your actual findings yield actual theoretical $\times 100$ what are the possible side reactions of nitration of bromobenzene dinitration can occur at ortho para or at both ortho positions occurs in overheating meta product happens very rarely because bromine is an ortho para director

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the soap in the lab is not suitable for this purpose 2 return to the lab and apply mineral oil to the affected area theory and explanation of the nitration of bromobenzene if the temperature of the reaction mixture exceeds 60 °C

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the bromine substituent in bromobenzene is an ortho para director the para position is favored due to steric effects in the reaction that takes place in this lab a nitro group is substituted onto bromobenzene at the para position to form 1-bromo-4-nitrobenzene

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study with quizlet and memorize flashcards containing terms like reacting sulfuric and nitric acid together produces the nitronium ion alone nitric acid is not a strong enough electrophile to react with bromobenzene sulfuric acid is added to the nitric acid the reverse rxn can overheat and boil easily includes a claisen adaptor and

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over 5 minutes slowly pipet this bromobenzene into the icy cold nitration mixture HNO_3 3 h 2 so 4 in the 125 ml flask and stir continue to stir the reaction mixture in the ice bath for 30 minutes do not heat in a new 250 ml erlenmeyer

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on the right the arenium cation has a nitro set with incredibly small 0.6 angstrom bond distances this will result in an optimization failure as the atoms are even close for the programme to accommodate their electrons demonstration of creating the p arenium cation by the nitration of bromobenzene

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the purpose of this lab was to observe where substituents are added on a benzene ring by way of electrophilic aromatic

substitution the electrophilic source in the nitration of bromobenzene comes from a nitronium ion NO_2^+ which acts as the electrophilic species

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nitration of bromobenzene is an electrophilic aromatic substitution reaction the electrophile will be formed from 4 0ml of nitric acid and 4 0ml of sulfuric acid and will then react with 4 5ml of bromobenzene this will produce o bromonitrobenzene and p bromonitrobenzene

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chem 2261 lab 5 nitration of bromobenzene introduction an electrophilic aromatic substitution reaction is an important process that allows for intermediate chemical structures to be used as precursors to pharmaceuticals and other industrially produced chemicals by adding different functional groups into an aromatic ring

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lab report 10 nitration of bromobenzene raekwon filmore cm 244 section 40 march 27 2018 introduction for this experiment nitration of bromobenzene was the focus of the lab the benzene is an aromatic compound and when it reacts with wither a mixture of sulfuric acid or nitric acid creates what is known as a nitro group

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