

### **Persepolis Questions**

*Answer these questions on a separate piece of paper with complete sentences.*

#### **Entry #1 Introduction – Chapter 4**

##### **Introduction**

1. According to the introduction, what stereotypical image is Satrapi trying to dispel?
2. The author indicates two motives for writing Persepolis. What are they?

##### **"The Veil" Chapter One**

1. What did the veil symbolize?
2. Why was a picture of Marjane's mother published in the newspaper? How did her mother respond? Why?
3. Describe the symbolism and conflict represented in the picture on the top of page 6.
4. Comment on the symbolism in the picture at the bottom of page 9. What three attributes of God does Marji wish to embody as a prophet?

##### **"The Bicycle" Chapter Two**

1. Explain the simile on page 10.
2. What metaphor does the author use on page 11 to portray the Persian people during 2500 years of tyranny and submission?
3. How is the picture on page 13 (middle) both symbolic and ironic?
4. Respond to the imagery on page 15.

##### **"The Water Cell" Chapter Three**

1. According to Marji, how was a king chosen? What was her rationale? What is her father's reaction? Why?
2. Comment on the image at the bottom of page 22. What change does it reflect?

##### **"Persepolis" Chapter Four**

1. What are the strengths and limitations of using the graphic cartoon to tell the story?
2. How is the image of the Shah on the bottom of page 27 both symbolic and ironic?
3. Ironically, what happens when a second dead man is carried out on a stretcher during a demonstration? What happens next that surprises Marji and amuses her family? What do the widow's actions show about the Persian people?

#### **Entry #2 – Chapter 5-Chapter 9**

##### **"The Letter" Chapter Five**

1. What happened on Black Friday? Why? What painful lesson did Marji learn?
2. Identify the irony and symbolism in the last picture on page 39.

##### **"The Party" Chapter Six**

1. Discuss an example of symbolism on page 40.
2. What story did Marji's neighbor make up about the spot on her cheek? What did this reflect about Iranian people?
3. Why does Ramin believe his father is not a murderer? How does his comment reflect the attitude of other children?

##### **"The Heroes" Chapter Seven**

1. What lessons does Marji learn about "truth"?
2. Comment on the images on page 51. What is Marji's response to what she hears?
3. According to Marji, what makes a hero?

##### **"Moscow" Chapter Eight**

1. What does the reader learn about the status of women in Iran?
2. How is the picture at the top of page 60 symbolic?

##### **"The Sheep" Chapter Nine**

1. What is the significance of the title? How does it help to characterize the people of Iran?
2. How does Satrapi carry the symbolism through this chapter?
3. Discuss how the artistic and literary elements on page 71 help to convey mood and theme?

#### **Entry #3 Chapter 10-Chapter 14**

##### **"The Trip" Chapter Ten**

1. How did the new Islamic republic impact the educational system?
2. How did the Fundamentalist woman and man differ from the modern woman and man?

# Persepolis Study Guide Questions

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web in today's experiment you will determine the solubility product ksp of calcium hydroxide  $\text{Ca(OH)}_2$  by measuring the concentration of  $\text{Ca(OH)}_2$  in a saturated solution calcium hydroxide is a sparingly soluble salt that dissolves according to the following reaction the solubility product expression for this reaction is

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web expert verified chem 182 determining the ksp of calcium hydroxide calcium hydroxide is an ionic solid that is sparingly soluble in water a saturated aqueous solution of  $\text{Ca(OH)}_2$  is represented in equation form as shown below  $\text{Ca(OH)}_2 \rightleftharpoons \text{Ca}^{2+} + 2\text{OH}^-$  the solubility product expression describes in mathematical terms the equilibrium

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web ksp for calcium hydroxide calcium hydroxide is a soft white caustic powder used in making mortar cements calcium salts paints and petrochemicals it is also used in saltwater aquaria to make up kalkwasser limewater solutions for reef tanks and is used as a ph regulating agent notice that calcium hydroxide is divalent and thus has twice

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web use the concentration of calcium hydroxide to calculate the concentration of calcium ions and hydroxide ions in the solution  $\text{Ca(OH)}_2 \rightleftharpoons \text{Ca}^{2+} + 2\text{OH}^-$   $5 \times 10^{-2} \text{ M}$  calcium ions  $5 \times 10^{-2} \text{ M}$   $2 \times 10^{-2} \text{ M}$  hydroxide ions calculate calcium hydroxide ksp using the concentrations of calcium ions and hydroxide ions

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web madison bowe elana keller april 15 2014 introduction the objective of this lab was to measure the  $K_{sp}$  of a slightly soluble basic salt and how solubility changes with temperature  $K_{sp}$  is another form of equilibrium expression that deals with solubility the reaction that will take place is  $\text{Ca(OH)}_2 \rightleftharpoons \text{Ca}^{2+} + 2\text{OH}^-$

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web question experiment 8 determining  $K_{sp}$  for calcium hydroxide post lab report sheet data table 2 points trial 1 trial 2  $\text{HCl}$  molarity 0.050 M volume of filtered  $\text{Ca(OH)}_2$  solution mL 5.00 5.00 3.71 13.75 equivalence point volume mL OH<sup>-</sup> in saturated  $\text{Ca(OH)}_2$  M  $\text{Ca}^{2+}$  in saturated  $\text{Ca(OH)}_2$  M  $K_{sp}$  of  $\text{Ca(OH)}_2$  average  $K_{sp}$  of  $\text{Ca(OH)}_2$

chemistry 1210 lab 11 experiment 11 determining the  $K_{sp}$  of calcium - Sep 08 2023

web the chemical equation used was  $\text{Ca(OH)}_2 \rightleftharpoons \text{Ca}^{2+} + 2\text{OH}^-$  the solubility product was determined to be  $K_{sp} \text{Ca}^{2+} \text{OH}^-$  from the titration which was made by dripping a concentrated solution of calcium hydroxide into hydrochloric acid the calcium hydroxide was filtered before starting the titration

18.1 solubility product constant  $K_{sp}$  chemistry libretexts - Feb 01 2023

web jul 12 2023 because the concentration of a pure solid such as  $\text{Ca}_3\text{PO}_4$  is a constant it does not appear explicitly in the equilibrium constant expression the equilibrium constant expression for the dissolution of calcium phosphate is therefore  $K_{sp} \text{Ca}_3\text{PO}_4 \rightleftharpoons 3\text{Ca}^{2+} + 4\text{PO}_4^{3-}$

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web lab report 8 determining  $K_{sp}$  of calcium hydroxide the purpose of this was to titrate a saturated  $\text{Ca(OH)}_2$  solution with a standard  $\text{HCl}$  solution and find the  $K_{sp}$  or the  $\text{OH}^-$  concentration was found and then the concentration was found and used for this experiment a titration of calcium hydroxide with hydrochloric acid

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web question determining the  $K_{sp}$  of calcium hydroxide  $\text{Ca(OH)}_2 \rightleftharpoons \text{Ca}^{2+} + 2\text{OH}^-$  for this experiment the titration of calcium hydroxide with hydrochloric acid was investigated determining the  $K_{sp}$  of calcium hydroxide  $\text{Ca(OH)}_2 \rightleftharpoons \text{Ca}^{2+} + 2\text{OH}^-$  for this experiment the titration of calcium

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web question chem 182 experiment 8 determining the  $K_{sp}$  of calcium hydroxide calcium hydroxide is an ionic solid that is sparingly soluble in water solution of  $\text{Ca(OH)}_2$  is represented in equation form solution of calcium hydroxide is an ionic solid that is sparingly soluble in water a saturated aqueous as shown below

determination of the  $K_{sp}$  of calcium hydroxide autosaved - Jun 05 2023

web therefore it was determined through error calculations and ease of use that the most reliable valid method to use to get

the closest  $K_{sp}$  value to the theoretical  $K_{sp}$  for  $Ca(OH)_2$  was the determination through pH method 2 introduction calcium hydroxide is a strong base soluble in water

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web plasma membrane homework symbiosis practice worksheet benzoin condensation grade a determining the  $K_{sp}$  of calcium hydroxide after reading the lab complete items a b c and d title purpose chemicals and equipment and summary of procedure as described on page 10 of exp 1 on an 8 1 2 x 11 sheet of paper

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web question lab 9 determination of the solubility product  $K_{sp}$  of calcium hydroxide introduction ionic compounds that are classified as insoluble based on solubility rules are actually slightly soluble each of these insoluble compounds actually dissolves to

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web we will filter the solid calcium hydroxide from the solution and then titrate the dissolved calcium hydroxide in solution with a known concentration of hydrochloric acid using titrimetric calculations we will determine the  $Ca^{2+}$  and  $OH^-$  concentrations in solution

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web calculate the  $K_{sp}$  for calcium hydroxide  $K_{sp} Ca(OH)_2 = 5.0 \times 10^{-6}$  find the accepted value of the  $K_{sp}$  for calcium hydroxide and compare it with your value discuss the discrepancy and suggest possible sources of experimental error accepted value for  $K_{sp}$  for sodium hydroxide is  $5.0 \times 10^{-6}$

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