

6**THE PERIODIC TABLE**

Chapter 6 Review Package - Part 1

A. Matching

Match each description in Column B with the correct term in Column A. Write the letter of the correct description on the line.

Column A

- _____ 1. metal
- _____ 2. halogen
- _____ 3. inner transition metal
- _____ 4. representative element
- _____ 5. ~~ionization energy~~
- _____ 6. alkaline earth metal
- _____ 7. cation
- _____ 8. electronegativity
- _____ 9. transition metal
- _____ 10. noble gas

Column B

- a. positively charged ion
- b. an element whose highest occupied *s* or *p* sublevels are partially filled
- c. ~~the energy needed to remove an electron from an atom in the gaseous state~~
- d. good conductor of heat and electric current
- e. an element whose highest occupied *s* sublevel and a nearby *d* sublevel contain electrons
- f. any nonmetal in Group 7A
- g. ~~the tendency of an atom to attract electrons when the atom is in a compound~~
- h. an element in which the highest occupied *s* and *p* sublevels are filled
- i. an element in Group 2A
- j. an element whose highest occupied *s* sublevel and a nearby *f* sublevel contain electrons

B. Multiple Choice

Choose the best answer and write its letter on the line.

- _____ 11. In the periodic table, there is a periodic pattern in the physical and chemical properties of elements when they are arranged in order of
 - a. increasing atomic mass.
 - b. increasing electronegativity.
 - c. increasing atomic radius.
 - d. increasing atomic number.

Periodic Trends Ch 6 Answers

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Periodic Trends Ch 6 Answers:

Unveiling the Magic of Words: A Overview of "**Periodic Trends Ch 6 Answers**"

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Table of Contents Periodic Trends Ch 6 Answers

1. Understanding the eBook Periodic Trends Ch 6 Answers
 - The Rise of Digital Reading Periodic Trends Ch 6 Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Periodic Trends Ch 6 Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Periodic Trends Ch 6 Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Periodic Trends Ch 6 Answers
 - Personalized Recommendations
 - Periodic Trends Ch 6 Answers User Reviews and Ratings
 - Periodic Trends Ch 6 Answers and Bestseller Lists

5. Accessing Periodic Trends Ch 6 Answers Free and Paid eBooks
 - Periodic Trends Ch 6 Answers Public Domain eBooks
 - Periodic Trends Ch 6 Answers eBook Subscription Services
 - Periodic Trends Ch 6 Answers Budget-Friendly Options
6. Navigating Periodic Trends Ch 6 Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Periodic Trends Ch 6 Answers Compatibility with Devices
 - Periodic Trends Ch 6 Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Periodic Trends Ch 6 Answers
 - Highlighting and Note-Taking Periodic Trends Ch 6 Answers
 - Interactive Elements Periodic Trends Ch 6 Answers
8. Staying Engaged with Periodic Trends Ch 6 Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Periodic Trends Ch 6 Answers
9. Balancing eBooks and Physical Books Periodic Trends Ch 6 Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Periodic Trends Ch 6 Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Periodic Trends Ch 6 Answers
 - Setting Reading Goals Periodic Trends Ch 6 Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Periodic Trends Ch 6 Answers
 - Fact-Checking eBook Content of Periodic Trends Ch 6 Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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