

**Directions:** Choose the letter that corresponds to the correct answer.

1. Which of the following can be considered as Dobereiner's triad?
  - a. He: 4.0, Li: 6.9, Na: 23.0
  - b. N: 14.0, P: 40.0, At: 217.0
  - c. Ne: 20.2, Ar: 39.9, Kr: 83.8
  - d. Ca: 40.1, Sr: 87.6, Ba: 137.3
2. Which of the following elements does not belong to the group?
  - a. N
  - b. S
  - c. Po
  - d. Te
3. Which of the following is/are TRUE about boron and oxygen?
  - a. both are metalloids
  - b. boron is smaller than oxygen
  - c. boron is more electronegative than oxygen
  - d. higher energy is required to eject an electron from the valence shell of boron
  - e. all of the above
  - f. none of the above

For 4 and 5, consider Na, Mg, Al, and Si.

4. The element that ejects one electron from its valence shell most readily is \_\_\_\_\_.
  - a. Na
  - b. Mg
  - c. Al
  - d. Si
5. Which element has the most negative electron affinity?
  - e. Na
  - f. Mg

# Periodic Trends Practice Questions Answers

**John T. Moore, Chris Hren, Peter J.  
Mikulecky**



## Periodic Trends Practice Questions Answers:

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specific topics Civil Mechanical Electrical Electronics Computer Science Study Strategies and Planning How to create a  
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### **Table of Contents Periodic Trends Practice Questions Answers**

1. Understanding the eBook Periodic Trends Practice Questions Answers
  - The Rise of Digital Reading Periodic Trends Practice Questions Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying Periodic Trends Practice Questions Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Periodic Trends Practice Questions Answers
  - User-Friendly Interface
4. Exploring eBook Recommendations from Periodic Trends Practice Questions Answers
  - Personalized Recommendations

- Periodic Trends Practice Questions Answers User Reviews and Ratings
- Periodic Trends Practice Questions Answers and Bestseller Lists
- 5. Accessing Periodic Trends Practice Questions Answers Free and Paid eBooks
  - Periodic Trends Practice Questions Answers Public Domain eBooks
  - Periodic Trends Practice Questions Answers eBook Subscription Services
  - Periodic Trends Practice Questions Answers Budget-Friendly Options
- 6. Navigating Periodic Trends Practice Questions Answers eBook Formats
  - ePub, PDF, MOBI, and More
  - Periodic Trends Practice Questions Answers Compatibility with Devices
  - Periodic Trends Practice Questions Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Periodic Trends Practice Questions Answers
  - Highlighting and Note-Taking Periodic Trends Practice Questions Answers
  - Interactive Elements Periodic Trends Practice Questions Answers
- 8. Staying Engaged with Periodic Trends Practice Questions Answers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Periodic Trends Practice Questions Answers
- 9. Balancing eBooks and Physical Books Periodic Trends Practice Questions Answers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Periodic Trends Practice Questions Answers
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Periodic Trends Practice Questions Answers
  - Setting Reading Goals Periodic Trends Practice Questions Answers
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Periodic Trends Practice Questions Answers
  - Fact-Checking eBook Content of Periodic Trends Practice Questions Answers

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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