

PERIODIC TABLE (BASIC)

1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18								
1 H Hydrogen 1.008	2 He Helium 4.003																								
3 Li Lithium 6.941	4 Be Beryllium 9.012																								
5 B Boron 10.81	6 C Carbon 12.01	7 N Nitrogen 14.01	8 O Oxygen 16.00	9 F Fluorine 18.99	10 Ne Neon 20.18																				
11 Na Sodium 22.99	12 Mg Magnesium 24.31																								
13 Al Aluminum 26.98	14 Si Silicon 28.09	15 P Phosphorus 30.97	16 S Sulfur 32.07	17 Cl Chlorine 35.45	18 Ar Argon 39.95																				
19 K Potassium 39.10	20 Ca Calcium 40.08																								
21 Sc Scandium 44.96	22 Ti Titanium 47.88	23 V Vanadium 50.94	24 Cr Chromium 52.00	25 Mn Manganese 54.94	26 Fe Iron 55.85	27 Co Cobalt 58.93	28 Ni Nickel 58.69	29 Cu Copper 63.55	30 Zn Zinc 65.38																
31 Ga Gallium 69.72	32 Ge Germanium 72.64	33 As Arsenic 74.92	34 Se Selenium 78.96	35 Br Bromine 79.90	36 Kr Krypton 83.80																				
37 Rb Rubidium 85.47	38 Sr Strontium 87.62																								
39 Y Yttrium 88.91	40 Zr Zirconium 91.22	41 Nb Niobium 92.91	42 Mo Molybdenum 95.94	43 Tc Technetium 98.91	44 Ru Ruthenium 101.07	45 Rh Rhodium 102.91	46 Pd Palladium 106.42	47 Ag Silver 107.87	48 Cd Cadmium 112.41	49 In Indium 114.82	50 Sn Tin 118.71	51 Sb Antimony 121.76	52 Te Tellurium 127.60	53 I Iodine 126.91	54 Xe Xenon 131.29										
55 Cs Cesium 132.91	56 Ba Barium 137.33																								
57 Pr Praseodymium 140.91	58 Nd Neodymium 144.24	59 Pm Promethium 144.91	60 Sm Samarium 150.36	61 Eu Europium 151.96	62 Gd Gadolinium 157.25	63 Tb Terbium 158.93	64 Dy Dysprosium 162.50	65 Ho Holmium 164.93	66 Er Erbium 167.26	67 Tm Thulium 168.93	68 Yb Ytterbium 173.05	69 Lu Lutetium 174.97													
71 La Lanthanum 138.91	72 Ce Cerium 140.12	73 Pr Praseodymium 140.91	74 Nd Neodymium 144.24	75 Pm Promethium 144.91	76 Sm Samarium 150.36	77 Eu Europium 151.96	78 Gd Gadolinium 157.25	79 Tb Terbium 158.93	80 Dy Dysprosium 162.50	81 Ho Holmium 164.93	82 Er Erbium 167.26	83 Tm Thulium 168.93	84 Yb Ytterbium 173.05	85 Lu Lutetium 174.97											
87 Fr Francium [223]	88 Ra Radium [226]																								

C Solid

Hg Liquid

H Gas

Rf Unknown

Metals

Other nonmetals

Nonmetals

Halogens

Noble gases

Metals

Alkali earth metals

Alkaline earth metals

Lanthanoids

Transition metals

Post-transition metals

Atomic #

Symbol

Element name

Weight

1

2

3

4

5

6

7

8

9

10

11

12

13

14

15

16

17

18

37°

For elements with no stable isotopes, the mass number of the isotope with the longest half-life is in parentheses.

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Hydrogen

Hydrogen is a chemical element with chemical symbol H and atomic number 1. With an atomic weight of 1.00794 u, hydrogen is the lightest element on the periodic table. Its monatomic form (H) is the most abundant chemical substance in the universe, constituting roughly 75% of all baryonic mass. Non-remnant stars are mainly composed of hydrogen in its plasma state. The most common isotope of hydrogen, termed protium (name rarely used, symbol 1H), has a single proton and zero neutrons.

Helium

Helium is a chemical element with symbol He and atomic number 2. It is a colorless, odorless, tasteless, non-toxic, inert, monatomic gas that heads the noble gas group in the periodic table. Its boiling and melting points are the lowest among the elements and it exists only as a gas except in extreme conditions.

Lithium

Lithium (from Greek: λίθος lithos, "stone") is a chemical element with symbol Li and atomic number 3. It is a soft, silver-white metal belonging to the alkali metal group of chemical elements. Under standard conditions it is the lightest metal and the least dense solid element. Like all alkali metals, lithium is highly reactive and flammable. For this reason, it is typically stored in mineral oil. When cut open, lithium exhibits a metallic luster, but contact with moist air corrodes the surface quickly to a dull silvery gray, then black tarnish. Because of its high reactivity, lithium never occurs freely in nature, and instead, only appears in compounds, which are usually ionic. Lithium occurs in a number of pegmatitic minerals, but due to its solubility as an ion is present in ocean water and is commonly obtained from brines and clays.



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