

Chapter 1: Limiting Reagents and Percent Yield (Pages 1-3)

Remember: Limiting reagent is the reactant that will completely react and leave another reactant

Excess: All reactants except the limiting reagent will have some left over. Remember: Limiting reagent always determines the amount of product.

Remember: Limiting reagent will determine how much product is produced and

byproducts will be. The limiting reagent is also a factor.



The ratio required for the composition for the limiting reagent.

2H ₂ (g)	100.00 percent
O ₂ (g)	50.00 percent
2H ₂ O(l)	100.00 percent
H ₂ O	100.00 percent

1. 100.00 g of each reactant will be used. 100.00 g of each reactant will be used. 100.00 g of each reactant will be used. 100.00 g of each reactant will be used. 100.00 g of each reactant will be used.



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4. 100.00 g of each reactant will be used. 100.00 g of each reactant will be used. 100.00 g of each reactant will be used. 100.00 g of each reactant will be used. 100.00 g of each reactant will be used.



Percent Yield And Limiting Reactants Quiz

Guillaume Favre



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