Question 5

10 points total Distribution of points

(a) (i) 2 points

From the ideal gas law

PV = nRT

For recognizing that $\frac{PV}{T}$ is constant throughout the cycle

Using the fact that pressure is the same for states 1 and 2

$$\frac{V_1}{T_1} = \frac{V_2}{T_2}$$

 $T_2 = \frac{V_2 T_1}{V_1}$

For substituting correct values into a correct expression

$$T_2 = (0.50 \text{ m}^3)(373 \text{ K})/(0.25 \text{ m}^3)$$

 $T_2 = 746 \text{ K}$

Note: Some students earned the first point by correctly calculating the value of n or the product nR using the given conditions in state 1. The student could have then proceeded to the correct substitutions in both parts (i) and (ii).

(iii) I point

Using the fact that volume is the same for states 1 and 3

$$\frac{P_1}{T_1} = \frac{P_3}{T_3}$$

$$T_3 = \frac{P_3 T_1}{P_1}$$

For substituting correct values into a correct expression

$$T_3 = (1.5 \times 10^5 \text{ Pa})(373 \text{ K})/(1.0 \times 10^5 \text{ Pa})$$

$$T_1 = 560 \text{ K}$$

I point

I point

I point

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