#### Question 5

10 points total Distribution of points

(a) (i) 2 points

From the ideal gas law

PV = nRT

For recognizing that  $\frac{PV}{T}$  is constant throughout the cycle

Using the fact that pressure is the same for states 1 and 2

$$\frac{V_1}{T_1} = \frac{V_2}{T_2}$$

 $T_2 = \frac{V_2 T_1}{V_1}$ 

For substituting correct values into a correct expression

$$T_2 = (0.50 \text{ m}^3)(373 \text{ K})/(0.25 \text{ m}^3)$$

 $T_2 = 746 \text{ K}$ 

Note: Some students earned the first point by correctly calculating the value of n or the product nR using the given conditions in state 1. The student could have then proceeded to the correct substitutions in both parts (i) and (ii).

(iii) I point

Using the fact that volume is the same for states 1 and 3

$$\frac{P_1}{T_1} = \frac{P_3}{T_3}$$

$$T_3 = \frac{P_3 T_1}{P_1}$$

For substituting correct values into a correct expression

$$T_3 = (1.5 \times 10^5 \text{ Pa})(373 \text{ K})/(1.0 \times 10^5 \text{ Pa})$$

$$T_1 = 560 \text{ K}$$

I point

I point

I point

# **Physics B 1995 Response Scoring Guidelines**

**Greg Jacobs, Joshua Schulman** 

## **Physics B 1995 Response Scoring Guidelines:**

Learning and Understanding National Research Council, Division of Behavioral and Social Sciences and Education, Center for Education, Committee on Programs for Advanced Study of Mathematics and Science in American High Schools, 2002-09-06 This book takes a fresh look at programs for advanced studies for high school students in the United States with a particular focus on the Advanced Placement and the International Baccalaureate programs and asks how advanced studies can be significantly improved in general It also examines two of the core issues surrounding these programs they can have a profound impact on other components of the education system and participation in the programs has become key to admission at selective institutions of higher education By looking at what could enhance the quality of high school advanced study programs as well as what precedes and comes after these programs this report provides teachers parents curriculum developers administrators college science and mathematics faculty and the educational research community with a detailed assessment that can be used to guide change within advanced study programs Introduction to Mario Molina Gilad James, PhD, Mario Molina is a world renowned scientist who is widely recognized for his groundbreaking research on the effects of man made chemicals on the environment Born in Mexico City in 1943 Molina was educated at the National Autonomous University of Mexico and went on to earn his PhD in chemistry from the University of California Berkeley He later taught at the Massachusetts Institute of Technology MIT and the University of California San Diego Molina is best known for his work on the depletion of the ozone layer a critical component of Earth's atmosphere that helps protect us from the harmful effects of ultraviolet radiation In 1995 he shared the Nobel Prize in Chemistry with two other scientists for their contributions to the understanding of the chemistry of the atmosphere Throughout his career Molina has been a tireless advocate for environmental protection He has worked tirelessly to raise awareness about the dangers of pollution and climate change and has been a vocal critic of policies that would undermine efforts to protect our planet In addition to his scientific work Molina has served in numerous advisory roles to governments and international organizations and has received numerous awards and accolades for his service to the scientific community His legacy as a pioneer in environmental science will continue to inspire future generations to work tirelessly to protect our planet for generations to come

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misunderstandings demonstrated by the TIMSS Advanced grade 12 students for these core concepts and shows how these can be traced back to poor foundational development of these concepts in earlier grades Patterns in misconceptions and misunderstandings are reported by grade country and gender In addition specific misconceptions and misunderstandings are tracked over time using trend items administered in multiple assessment cycles The study and associated methodology may enable education systems to help identify specific needs in the curriculum improve inform instruction across grades and also raise possibilities for future TIMSS assessment design and reporting that may provide more diagnostic outcomes

Stochastic and Dynamic Views of Chemical Reaction Kinetics in Solutions A. M. Kuznet□s□ov (dokt. fiz.-mat. nauk.),Alexander M. Kuznetsov,1999-01-01

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