

QUESTION 1

1.1 Intermolecular forces are attractive forces between... (2)

- A. protons and electrons
- B. atoms
- C. molecules
- D. substances

1.2 What is the name of the polymer made of the monomer:
 $\text{CH}_2 = \text{CH}_2$ (2)

- A. Ethene
- B. Polyethene
- C. Polyester
- D. PVC

1.3 The following reaction occurs:



Which of the following can be used to measure the rate of this reaction? (2)

- A. Measuring the volume of gas produced per unit time
- B. Observing the precipitate formed
- C. Observing the changes in mass
- D. All of the above

1.4 The reaction below has reached equilibrium at a temperature of 50°C in a sealed gas syringe.



Consider the following statements:

- I. An increase in the amount of H_2S will favor the forward reaction and increase the K_c value
- II. An increase in pressure will not change the K_c -value and will not favor the forward or the reverse reaction.
- III. An increase in temperature will favor the forward reaction and

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RM Cervero



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