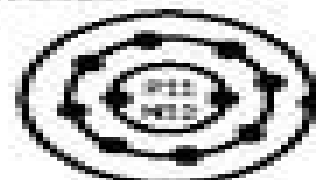


Semester I Final Exam: Physical Science

Name: _____ Date: _____ Period: _____

Mark the correct answer or on your answer document.



1. What is the **atomic number** for the element above?

- a) 24
- b) 12
- c) 24.305
- d) Cannot be determined

2. What is the **atomic weight** for the element above?

- a) 24
- b) 12
- c) 24.305
- d) Cannot be determined

3. What is the **mass number** for the element above?

- a) 24
- b) 12
- c) 24.305
- d) Cannot be determined

4. What is the **charge** on an electron?

- a) negative
- b) positive
- c) neutral
- d) Cannot be determined

5. The number of **protons** in the nucleus of an atom determine the?

- a) Mass of the element
- b) The charge on the element
- c) The type of element
- d) Whether it is radioactive or not

6. Which **element** is being represented by the atom above?

- a) Mg (Magnesium)
- b) Na (Sodium)
- c) V (Vanadium)
- d) Na (Sodium)

7. What is the **outer electron** in the outer shell called?

- a) Single shell electron
- b) Valence electron
- c) Electron
- d) Neutral atom electron

8. What is the **mass number** for this atom?

- a) 20
- b) 22.9
- c) 11
- d) 12

9. The **atomic weight** is the average weight of all?

- a) Isotopes of that atom
- b) Protons in that atom
- c) Protons and electrons in that atom
- d) Neutrons in that atom

10. The periodic table of elements is **arranged in**.

- a) Atomic mass
- b) Atomic number
- c) Isotopes
- d) Numbers of electrons

Physical Science Semester B Exam Answers

Greg Jacobs, Joshua Schulman



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Table of Contents Physical Science Semester B Exam Answers

1. Understanding the eBook Physical Science Semester B Exam Answers
 - The Rise of Digital Reading Physical Science Semester B Exam Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Physical Science Semester B Exam Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Physical Science Semester B Exam Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Physical Science Semester B Exam Answers
 - Personalized Recommendations
 - Physical Science Semester B Exam Answers User Reviews and Ratings
 - Physical Science Semester B Exam Answers and Bestseller Lists
5. Accessing Physical Science Semester B Exam Answers Free and Paid eBooks
 - Physical Science Semester B Exam Answers Public Domain eBooks
 - Physical Science Semester B Exam Answers eBook Subscription Services
 - Physical Science Semester B Exam Answers Budget-Friendly Options
6. Navigating Physical Science Semester B Exam Answers eBook Formats

- ePub, PDF, MOBI, and More
- Physical Science Semester B Exam Answers Compatibility with Devices
- Physical Science Semester B Exam Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Physical Science Semester B Exam Answers
 - Highlighting and Note-Taking Physical Science Semester B Exam Answers
 - Interactive Elements Physical Science Semester B Exam Answers
- 8. Staying Engaged with Physical Science Semester B Exam Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Physical Science Semester B Exam Answers
- 9. Balancing eBooks and Physical Books Physical Science Semester B Exam Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Physical Science Semester B Exam Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Physical Science Semester B Exam Answers
 - Setting Reading Goals Physical Science Semester B Exam Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Physical Science Semester B Exam Answers
 - Fact-Checking eBook Content of Physical Science Semester B Exam Answers
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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