

Answer KEY

Unit 3 Outline/ Study Guide - Electron Configuration and Periodic Trends

Essential Skills/ State Standards:

1. Students know how to relate the position of an element in the periodic table to its quantum electron configuration and to its reactivity with other elements in the table.
2. Students know how to use the periodic table to identify metals, semimetals, nonmetals, halogens, alkali metals, alkaline earth metals, and transition metals.
3. Students know how to use the periodic table to identify the following trends:
 ionization energy, electronegativity, and the relative sizes of ions and atoms
 (this includes knowing the definitions of ionization energy, electronegativity, cation, and anion.)

Study Guide - (use your notes and worksheets for a more complete review)

1. Identify the following things on the periodic table:

-metals	-alkali metals
-metalloids (semi-metals)	-alkaline earth metals
-transition metals	-halogens
-nonmetals	-noble gases
-Periods	-Groups/ families
Identify groups with the following numbers of valence electrons: 1, 2, 3, 4, 5, 6, 7, 8	

2. Why do the elements in the groups (columns) behave so similar to each other even if they have great differences in their number of protons? **same number of valence electrons and similar chemical properties.**
3. What is the connection between the following terms: **octet rule, valence electrons, and ions?**
In order to achieve the octet rule, atoms gain or lose their valence electrons. They become charged ions when this occurs.
4. a. List the elements that are exceptions to the octet rule b) Why don't these atoms achieve the octet rule?
 a. Small atoms are happy with 2 electrons (ex: H, He, Li, Be, B)
 b. because the 1s energy level only holds 2 electrons

Periodic Trends Test Review Answers

The Princeton Review



Periodic Trends Test Review Answers:

This book delves into Periodic Trends Test Review Answers. Periodic Trends Test Review Answers is a crucial topic that needs to be grasped by everyone, ranging from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Periodic Trends Test Review Answers, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:
 - Chapter 1: Introduction to Periodic Trends Test Review Answers
 - Chapter 2: Essential Elements of Periodic Trends Test Review Answers
 - Chapter 3: Periodic Trends Test Review Answers in Everyday Life
 - Chapter 4: Periodic Trends Test Review Answers in Specific Contexts
 - Chapter 5: Conclusion
2. In chapter 1, this book will provide an overview of Periodic Trends Test Review Answers. The first chapter will explore what Periodic Trends Test Review Answers is, why Periodic Trends Test Review Answers is vital, and how to effectively learn about Periodic Trends Test Review Answers.
3. In chapter 2, the author will delve into the foundational concepts of Periodic Trends Test Review Answers. This chapter will elucidate the essential principles that must be understood to grasp Periodic Trends Test Review Answers in its entirety.
4. In chapter 3, this book will examine the practical applications of Periodic Trends Test Review Answers in daily life. This chapter will showcase real-world examples of how Periodic Trends Test Review Answers can be effectively utilized in everyday scenarios.
5. In chapter 4, the author will scrutinize the relevance of Periodic Trends Test Review Answers in specific contexts. This chapter will explore how Periodic Trends Test Review Answers is applied in specialized fields, such as education, business, and technology.
6. In chapter 5, the author will draw a conclusion about Periodic Trends Test Review Answers. The final chapter will summarize the key points that have been discussed throughout the book.

The book is crafted in an easy-to-understand language and is complemented by engaging illustrations. It is highly recommended for anyone seeking to gain a comprehensive understanding of Periodic Trends Test Review Answers.

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