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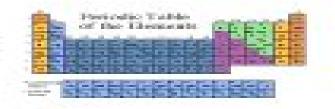
Ferrings

## Periodic Table WebQuest

## Instructions: In each of the websites listed, find the information asked for.

## Part I: "Get Organized Periodically"

https://www.chendidds.com/files/elem-netable.html



- L. Why are the elements placed in specific places on the Periodic Table?
- Pertods are \_\_\_\_\_\_\_that run from left to right.
- Elements in the same period have the same \_\_\_\_\_\_
- 4. Every element to the first period has shell for its . Every element in the second period has fee its . See the pattern?
- Groups are \_\_\_\_\_\_\_ that run from top to bottom.
- Every element in group one has \_\_\_\_\_\_\_ electron to its outer shell. Every element in group two has
  electrons in its outer shell.
- 8. Hydrogen is special because it can are like two groups, \_\_\_\_\_ and \_\_\_\_\_.
- 10. Although helium has only \_\_\_\_\_\_ electrons in its outer shell, it is grouped with elements that have
- 11. The green elements on this table are called \_\_\_\_\_\_\_\_ elements. They each have two electrons to their outer shell.

#### Peri II: "Family Fun" Go to http://chestealchestests.com/

- 12. Click on Alkali Metals (left bir) and answer the following questions.
  - a. What is the group number?
  - b. Are these mesals reactive?
  - c. Do these metals occur freely in nature?
  - d. How many electrons are in their outer shell?

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**S Ben Porath** 

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