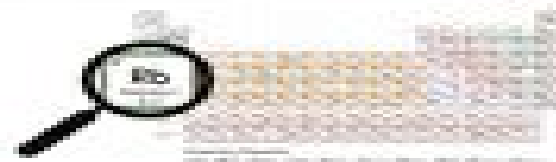


Periodic Table: Scavenger Hunt



Read the information below and answer the questions using the periodic table on page 2.

The **periodic table** is a graphical representation of all known chemical elements. Each box on the periodic table contains information about an element. The labeled image below shows the box on the periodic table for the element hydrogen.



The elements are ordered by their atomic number. Elements in the periodic table are also grouped by their chemical properties. The periodic table is color coded based on these properties.

Use the periodic table on page 2 to find the element or elements that match each description.

1. The **alkali metal** with the atomic number 19. _____
2. The **nonmetal** with an atomic mass of 32.06 _____
3. The **other metal** with an atomic mass of 26.98 _____
4. The **metalloid** with the atomic number 32 _____
5. The **noble gas** with the smallest atomic mass _____
6. The **transition metal** with the chemical symbol Au _____
7. The **lanthanide** with the atomic number 60 _____
8. The **actinide** with an atomic mass of 232.04 _____
9. The **transition metal** with the smallest atomic mass and an atomic number between 27 and 30 _____
10. The two **alkaline earth metals** with chemical symbols that start with the letter "B"
a. _____ b. _____

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Differentiation That Really Works Cheryll M. Adams, Rebecca L. Pierce, 2021-09-23 Differentiation That Really Works Science provides time saving tips and strategies from real teachers who teach science in grades 6-12. These teachers not only developed the materials and used them in their own classes but they also provided useful feedback and comments about the activities. The strategies included in the book are: tiered lessons, cubing, graphic organizers, exit cards, learning contracts, and choice boards. Every strategy includes directions and offers opportunities for differentiation. Grades 6-12. *Current Index to Journals in Education*, 1985. The Software Encyclopedia 2001, 2001. **Periodic Table Advanced** BarCharts, Inc., Mark Jackson, 2014-12-31. The ultimate reference tool and lab partner for any student of science. Durable, laminated, and designed to fit as much info as possible in this handy 6-page format. Separate property tables are broken out for the ease of locating trends while studying and working while other pages offer essential notes about the table's organization and history. Consistently a best seller since its first creation. *Periodic Table, The: Past, Present, And Future* Geoff Rayner-canham, 2020-08-04. This is an absolutely wonderful book that is full of gems about the elements and the periodic table. All in all, the book is highly recommended to philosophers of chemistry. As philosophers, we have a natural tendency to concentrate on generalities and not to get too involved in the specifics and the details. Above all else, this new book reminds us that such an approach needs to be tempered by a detailed knowledge of the exceptions and features that go against the simplified generalities which we so cherish. Read Full Review. *Eric Scerri Foundations of Chemistry*. Many questions are dealt with in a clearly written way in this stimulating and innovative book. The reader will quickly become interested in the subject and will be taken on tour through this Periodic Table in a very readable way, both for students and teachers. The number of illustrations is good and clear. This book is indeed unique and quite thought provoking. This book is highly recommended for students, teachers, researchers, and not only chemists. Geologists, biochemists, and also physicists will find it very interesting to read. Read Full Review. *Chemistry International*. That fossilized chart on every classroom wall isn't that The Periodic Table. Isn't that what Mendeleev devised about a century ago? No and No. There are many ways of organizing the chemical elements, some of which are thought provoking and which reveal philosophical challenges. Where does hydrogen belong? Can an element occupy more than one location on the chart? Which are the Group 3 elements? Is aluminum in the wrong place? Why is silver I like thallium I? Why is vanadium like molybdenum? Why does gold form an auride ion like a halide ion? Does an atom know if it is a non-metal or metal? Which elements are the metalloids? Which are the triels? So many questions. In this stimulating and innovative book, the Reader will be taken on a voyage from the past to the present to the future of the Periodic Table. This book is unique. This book is readable. This book is thought provoking. It is a multi-dimensional examination of patterns and trends among the chemical elements. Every reader will discover something about the chemical elements which will provoke thought and a new appreciation as to how the elements relate together. Chemistry and Our Universe,

Return to the periodic table introduced in Lecture 1 to practice predicting properties of elements based on their electronic structure Then witness what happens when three different alkali metals react with water Theory forecasts a pronounced difference in the result Is there

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Science Comics: The Periodic Table of Elements Jon Chad,2023-03-14 Step out of your element with Science Comics The Periodic Table of Elements the latest volume of First Second s nonfiction graphic novel series A book of fun chemistry experiments has fallen into the wrong hands Only Mel can use her knowledge of the periodic table to put an end to a maniacal madman s evil schemes The periodic table helps us quickly understand the 118 elements those tiny substances that make up everything in the world By using the periodic table we can recognize how these building blocks behave find trends and patterns in the universe and make predictions about elements that haven t been discovered yet Join us in learning about the periodic table and maybe the next big discovery will be yours

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com *The Periodic Table* Eric R. Scerri, 2019-07-25 The periodic table of elements first encountered by many of us at school provides an arrangement of the chemical elements ordered by their atomic number electron configuration and recurring chemical properties and divided into periodic trends In this Very Short Introduction Eric R Scerri looks at the trends in properties of elements that led to the construction of the table and shows how the deeper meaning of the table's structure gradually became apparent with the development of atomic theory and in particular quantum mechanics which underlies the behaviour of all of the elements and their compounds This new edition publishing in the International Year of the Periodic Table celebrates the completion of the seventh period of the table with the ratification and naming of elements 113 115 117 and 118 as nihonium moscovium tennessine and oganesson Eric R Scerri also incorporates new material on recent advances in our understanding of the origin of the elements as well as developments concerning group three of the periodic table ABOUT THE SERIES The Very Short Introductions series from Oxford University Press contains hundreds of titles in almost every subject area These pocket sized books are the perfect way to get ahead in a new subject quickly Our expert authors combine facts analysis perspective new ideas and enthusiasm to make interesting and challenging topics highly readable **Draw the Periodic Table of the Elements from Memory** Eric Fontaine, **Mastering the Periodic Table** Linda Trombley, 2000 *The Periodic Table* Tom Jackson, 2020-02-04 Which is the densest element Which has the largest atoms And why are some elements radioactive From the little known uses of gold in medicine to the development of the hydrogen bomb this is a fresh new look at the Periodic Table Combining cutting edge science with fascinating facts and stunning infographics this book looks at the extraordinary stories of discovery amazing properties and surprising uses of each elements whether solid liquid or gas naturally occurring synthesised or theoretical From hydrogen to oganesson this is a fact filled visual guide to each element each accompanied by technical data category atomic number weight boiling point as well as fun facts and stories about their discovery and surprising uses **Periodic Table Advanced** BarCharts, Inc, 2014

The Periodic Table Eric R. Scerri, 2006-10-12 The periodic table is one of the most potent icons in science It lies at the core of chemistry and embodies the most fundamental principles of the field The one definitive text on the development of the periodic table by van Spronsen 1969 has been out of print for a considerable time The present book provides a successor to van Spronsen but goes further in giving an evaluation of the extent to which modern physics has or has not explained the periodic system The book is written in a lively style to appeal to experts and interested lay persons alike The Periodic Table begins with an overview of the importance of the periodic table and of the elements and it examines the manner in which the term element has been interpreted by chemists and philosophers The book then turns to a systematic account of the early developments that led to the classification of the elements including the work of Lavoisier Boyle and Dalton and Cannizzaro The precursors to the periodic system like D bereiner and Gmelin are discussed In chapter 3 the discovery of the periodic system by six independent scientists is examined in detail Two chapters are devoted to the discoveries of Mendeleev the

leading discoverer including his predictions of new elements and his accommodation of already existing elements Chapters 6 and 7 consider the impact of physics including the discoveries of radioactivity and isotopy and successive theories of the electron including Bohr's quantum theoretical approach Chapter 8 discusses the response to the new physical theories by chemists such as Lewis and Bury who were able to draw on detailed chemical knowledge to correct some of the early electronic configurations published by Bohr and others Chapter 9 provides a critical analysis of the extent to which modern quantum mechanics is or is not able to explain the periodic system from first principles Finally chapter 10 considers the way that the elements evolved following the Big Bang and in the interior of stars The book closes with an examination of further chemical aspects including lesser known trends within the periodic system such as the knight's move relationship and secondary periodicity as well as attempts to explain such trends

The Periodic Table DK, 2025-09-04 This is the periodic table as you've never seen it before By presenting the elements in the order in which they were discovered The Periodic Table also tells the story of chemistry Now iconic the table was invented in the 1800s long before anyone knew what was inside an atom As more elements were found the table got bigger Some were discovered at risk to life and limb for example the hunt for the explosive element fluorine injured or even killed several scientists who came to be known as the fluorine martyrs Even now the table is still growing as scientists push the boundaries and add to the 118 elements discovered so far Some of the newest elements have been observed only as they briefly flicker into existence in the extreme conditions inside particle colliders and accelerators Gorgeous photographs show even the most familiar elements in a new light Bold clear infographics and artworks reveal aspects of their atomic structure that explain why they behave the way they do For example why is gold so malleable and diamond so incredibly hard How can bismuth levitate and why does cobalt make paint such a vivid blue If you want to unlock the secrets of matter itself open The Periodic Table

The Periodic Table Adrian Dingle, Dan Green, 2014 The original Basher Science made even better

Trends in the Periodic Table Open University. S25- Course Team, 1972

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