

Periodic Table of the Elements

KEY

Atomic Mass → 12.011 ← Selected Oxidation States
Symbol → C
Atomic Number → 6
Electron Configuration → 2-4

Relative atomic masses are based on $^{12}\text{C} = 12$ (exact)
Notes: Numbers in parentheses are mass numbers of the most stable or common isotope.

1	2											13	14	15	16	17	18																												
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4 19 K 39.10	20 40 Ca 40.08	21 44.96 Sc	22 47.88 Ti	23 50.94 V	24 51.99 Cr	25 54.94 Mn	26 55.85 Fe	27 58.93 Co	28 58.93 Ni	29 63.55 Cu	30 65.38 Zn	31 69.72 Ga	32 72.64 Ge	33 74.92 As	34 78.96 Se	35 79.90 Br	36 83.80 Kr																												
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Period	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18
1	H	He																
2	Li	Be											B	C	N	O	F	Ne
3	Na	Mg											Al	Si	P	S	Cl	Ar
4	K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr
5	Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I	Xe
6	Cs	Ba	La	Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At	Rn
7	Fr	Ra	Ac	Rf	Db	Sg	Bh	Hs	Mt	Ds	Rg	Cn	Uut	Uuq	Uup	Uuh	Uus	Uuo

Lanthanide Series: Ce, Pr, Nd, Pm, Sm, Eu, Gd, Tb, Dy, Ho, Er, Tm, Yb, Lu

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RC Schank



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