

# PERIODIC TABLE (ADVANCED)

## Molecular Structure of Hydrogen



### Basic Information

Name: Hydrogen  
Symbol: H  
Atomic Number: 1  
Atomic Mass: 1.00794 amu  
Melting Point:  $-259.14^{\circ}\text{C}$  ( $-314.009165^{\circ}\text{K}$ ,  $-434.41203^{\circ}\text{F}$ )  
Boiling Point:  $-252.87^{\circ}\text{C}$  ( $-29.280005^{\circ}\text{K}$ ,  $-423.166^{\circ}\text{F}$ )  
Number of Protons/Electrons: 1  
Number of Neutrons: 0  
Classification: Non-metal  
Crystal Structure: Hexagonal  
Density @ 293 K: 0.08989 g/cm<sup>3</sup>  
Color: colorless

### Atomic Structure

Number of Energy Levels: 1  
First Energy Level: 1

### Isotopes

| Isotope | Half-life  |
|---------|------------|
| H-1     | Stable     |
| H-2     | Stable     |
| H-3     | 12.3 years |

### Facts

Name Origin: from words *hydro* (water) and *genes* (generated)  
Uses: balloons, metal refining  
Obtained From: mines, oil, gas wells

## Molecular Structure of Carbon



### Basic Information

Name: Carbon  
Symbol: C  
Atomic Number: 6  
Atomic Mass: 12.0107 amu  
Melting Point:  $3500.0^{\circ}\text{C}$  ( $3773.15^{\circ}\text{K}$ ,  $6532.8^{\circ}\text{F}$ )  
Boiling Point:  $4827.8^{\circ}\text{C}$  ( $5100.95^{\circ}\text{K}$ ,  $8700.0^{\circ}\text{F}$ )  
Number of Protons/Electrons: 6  
Number of Neutrons: 6  
Classification: Non-metal  
Crystal Structure: Hexagonal  
Density @ 293 K: 2.62 g/cm<sup>3</sup>  
Color: May be black

### Atomic Structure

Number of Energy Levels: 2  
First Energy Level: 2      Second Energy Level: 4

### Isotopes

| Isotope | Half-life    |
|---------|--------------|
| C-11    | 20.3 minutes |
| C-12    | Stable       |
| C-13    | Stable       |
| C-14    | 5730.0 years |
| C-15    | 2.5 seconds  |

### Facts

Uses: steel, fibers  
Obtained From: burning with insufficient oxygen

# Periodic Table Advanced Speedy Study Guide

**R Sanford**



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