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Exceed to Succeed

Worksheet: Periodic Trends

1. ATOMIC RADIUS

For each of the following sets of atoms, rank the atoms from smallest to largest atomic radius.

- a. Li, C, F
- b. Li, Na, K
- c. Ge, P, O
- d. C, N, Al
- e. Al, Cl, Ga

2. IONIC RADIUS

For each of the following sets of ions, rank them from smallest to largest ionic radius.

- a. Mg^{2+} , Si^{4+} , S^{2-}
- b. Mg^{2+} , Ca^{2+} , Ba^{2+}
- c. F^- , Cl^- , Br^-
- d. Ba^{2+} , Cu^{2+} , Zn^{2+}
- e. Si^{4+} , P^{3-} , O^{2-}

3. IONIZATION ENERGY

For each of the following sets of atoms, rank them from lowest to highest ionization energy.

- a. Mg, Si, S
- b. Mg, Ca, Ba
- c. F, Cl, Br
- d. Ba, Cu, Ne
- e. Si, P, He

4. ELECTRONEGATIVITY

For each of the following sets of atoms, rank them from lowest to highest electronegativity.

- a. Li, C, N
- b. C, O, Ne
- c. Si, P, O
- d. K, Mg, P
- e. S, F, He

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CH Cherryholmes



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Table of Contents Periodic Trends Chapter 5 Answer Key

1. Understanding the eBook Periodic Trends Chapter 5 Answer Key
 - The Rise of Digital Reading Periodic Trends Chapter 5 Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Periodic Trends Chapter 5 Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Periodic Trends Chapter 5 Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Periodic Trends Chapter 5 Answer Key
 - Personalized Recommendations
 - Periodic Trends Chapter 5 Answer Key User Reviews and Ratings
 - Periodic Trends Chapter 5 Answer Key and Bestseller Lists

5. Accessing Periodic Trends Chapter 5 Answer Key Free and Paid eBooks
 - Periodic Trends Chapter 5 Answer Key Public Domain eBooks
 - Periodic Trends Chapter 5 Answer Key eBook Subscription Services
 - Periodic Trends Chapter 5 Answer Key Budget-Friendly Options
6. Navigating Periodic Trends Chapter 5 Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Periodic Trends Chapter 5 Answer Key Compatibility with Devices
 - Periodic Trends Chapter 5 Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Periodic Trends Chapter 5 Answer Key
 - Highlighting and Note-Taking Periodic Trends Chapter 5 Answer Key
 - Interactive Elements Periodic Trends Chapter 5 Answer Key
8. Staying Engaged with Periodic Trends Chapter 5 Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Periodic Trends Chapter 5 Answer Key
9. Balancing eBooks and Physical Books Periodic Trends Chapter 5 Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Periodic Trends Chapter 5 Answer Key
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Periodic Trends Chapter 5 Answer Key
 - Setting Reading Goals Periodic Trends Chapter 5 Answer Key
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Periodic Trends Chapter 5 Answer Key
 - Fact-Checking eBook Content of Periodic Trends Chapter 5 Answer Key
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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