

Graphing Periodic Trends Activity

Name: _____ Period: _____

Use the information in these tables to complete the graph as described below

Symbol	Atomic Radius (Picometers)	First Ionization energy (kilojoules/mole)	Electronegativity (4-point scale)
H	31	1312	2.1
Li	128	520	1.0
Na	166	496	0.9
K	203	410	0.8
Rb	220	403	0.8
Cs	244	376	0.7

Symbol	Atomic Radius (Picometers)	First Ionization energy (kilojoules/mole)	Electronegativity (4-point scale)
Na	166	496	0.9
Mg	141	738	1.2
Al	121	578	1.5
Si	111	787	1.8
P	107	1012	2.1
S	105	1000	2.5
Cl	102	1251	3.0
Ar	106	1521	---

1. Using colored pencils or pen, list under the symbol (in this order!) the Atomic Radius, First Ionization Energy, and Electronegativity.
2. Use a different color for each property. Example: Write all of the Atomic radius values in red, all of the First ionization energies in green, and all of the Electronegativities in blue. Of course, you can pick any colors that are available, as long as you are consistent.
3. Observe the trends in each property as you go down the Alkali metal group, and as you go across Period 3.
4. Write out each of the statements written on the opposite side, completing each statement with the observed trend (increase or decrease)

H	<p>Across a period (→) atomic radius tends to _____.</p> <p>Across a period (→) first ionization energy tends to _____.</p> <p>Across a period (→) electronegativity tends to _____.</p>						
Li							
Na 166 496 0.9	Mg	Al	Si	P	S	Cl	Ar
K	<p>Down a Group (↓) atomic radius tends to _____.</p> <p>Down a Group (↓) first ionization energy tends to _____.</p> <p>Down a Group (↓) electronegativity tends to _____.</p>						
Rb							
Cs							

Periodic Trends A Graphical Analysis Answer Key

Jicheng Xie



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ChemDiscovery Student Guide Olga I. Agapova, 2002 Bulletin - SLA Geography and Map Division Special Libraries Association. Geography and Map Division, 1968 U.S. Geological Survey Water-supply Paper, 1982 **Geological Survey Water-supply Paper**, 1984 *Census Education Project*, 1990, 1988 Prentice Hall Chemistry, 2000 Computability Theory and Its Applications Peter Cholak, 2000 This collection of articles presents a snapshot of the status of computability theory at the end of the millennium and a list of fruitful directions for future research The papers represent the works of experts in the field who were invited speakers at the AMS IMS SIAM 1999 Summer Conference on Computability Theory and Applications which focused on open problems in computability theory and on some related areas in which the ideas methods and or results of computability theory play a role Some presentations are narrowly focused others cover a wider area Topics included from pure computability theory are the computably enumerable degrees M Lerman the computably enumerable sets P Cholak R Soare definability issues in the c e and Turing degrees A Nies R Shore and other degree structures M Arslanov S Badaev and S Goncharov P Odifreddi A Sorbi The topics involving relations between computability and other areas of logic and mathematics are reverse mathematics and proof theory D Cenzer and C Jockusch C Chong and Y Yang H Friedman and S Simpson set theory R Dougherty and A Kechris M Groszek T Slaman and computable mathematics and model theory K Ambos Spies and A Kucera R Downey and J Remmel S Goncharov and B Khoussainov J Knight M Peretyatkin A Shlapentokh

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