



Percent Yield

45 grams of nitrogen monoxide (NO) react with an excess of sodium hydroxide (NaOH) following the balanced equation above. 6 grams of water were obtained from the reaction

What is the % yield?

Nitrogen monoxide (NO) gmm:

$$(1)(14.0) + (1)(16.0) = 30.0 \text{ g/mol}$$

$$\text{Actual Yield} = 6 \text{ g H}_2\text{O}$$

Water (H₂O) gmm:

$$(2)(1.0) + (1)(16.0) = 18.0 \text{ g/mol}$$

$$\text{Theoretical Yield} = 6.75 \text{ g H}_2\text{O}$$

$$\frac{\text{Actual Yield}}{\text{Theoretical Yield}} \times 100 = 88.9\%$$

$$\frac{45 \text{ g NO} \times \frac{1 \text{ mol NO}}{30 \text{ g NO}} \times \frac{4 \text{ mol NO}}{4 \text{ mol NO}} \times \frac{1 \text{ mol H}_2\text{O}}{1 \text{ mol H}_2\text{O}} \times 18 \text{ g H}_2\text{O}}{1} = 6.75 \text{ g H}_2\text{O}$$

Percent Yield Problems And Answers

S Ben Porath

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

Percent Yield Problems And Answers:

Embracing the Song of Appearance: An Emotional Symphony within **Percent Yield Problems And Answers**

In a global taken by screens and the ceaseless chatter of quick interaction, the melodic beauty and emotional symphony created by the published term frequently fade in to the background, eclipsed by the persistent sound and interruptions that permeate our lives. But, located within the pages of **Percent Yield Problems And Answers** a wonderful literary value brimming with raw emotions, lies an immersive symphony waiting to be embraced. Constructed by a masterful composer of language, that charming masterpiece conducts viewers on an emotional trip, skillfully unraveling the concealed songs and profound impact resonating within each carefully crafted phrase. Within the depths of the poignant review, we shall examine the book's central harmonies, analyze its enthralling writing fashion, and surrender ourselves to the profound resonance that echoes in the depths of readers' souls.

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