

Instructions:-

Question paper consist of 31 questions divided into four sections namely A, B, C, D.

- 1) **Section-A:** Q. No. 1 contains ten multiple choice types of questions carrying one mark each.
Q.No.2 contains eight very short answer types of questions carrying one mark each.
- 2) **Section-B :** Q.No.3 to Q. No. 14 are Twelve questions carrying two mark each out of which any eight to be attempted.
- 3) **Section-C :** Q. No.15 to Q.No.26 are Twelve questions carrying three mark each out of which any eight are to be attempted.
- 4) **Section-D :** Q.No.27 to Q.No. 31 are Five questions carrying four mark each out of which any three to be attempted.

Section- A

Q.1) Select and write the correct answer:

(10)

- i. Which of the following compounds can NOT demonstrate the law of proportions?
a) NO, NO₂ b) CO, CO₂ c) H₂O, H₂O₂ d) Na₂S, NaF
- ii. Which of the following temperature will read the same value on Celsius and Fahrenheit Scales
a) -40 b) + 40 c) -80 d) -20
- iii. SI unit of the quantity electric current is
a) Volts b) Ampere c) Candela d) Newton
- iv. The no. of molecules in 22.4 cm³ of Nitrogen Gas at STP is
a) 6.022×10^{20} b) 6.022×10^{23} c) 22.4×10^{20} d) 22.4×10^{23}
- v. The homogeneous series of alcohol has general molecular formula ?
a) C_n H₂n+2OH b) C_n H₂n OH c) C_n H₂n+1OH d) none
- vi. Find the molar mass of CH₃CHO.
a) 44 b) 40 c) 80 d) 30
- vii. Find the molar mass of CH₃NO₂.
a) 61 b) 77 c) 98 d) 51
- viii. Atomic number of an elements is equal to
a) P = e- b) p = N c) P = A d) p = n = e

Nmaths 2examination Paper

Peter Sheahan, Liz Ramsay



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Table of Contents Nmaths 2examination Paper

1. Understanding the eBook Nmaths 2examination Paper
 - The Rise of Digital Reading Nmaths 2examination Paper
 - Advantages of eBooks Over Traditional Books
2. Identifying Nmaths 2examination Paper
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Nmaths 2examination Paper
 - User-Friendly Interface
4. Exploring eBook Recommendations from Nmaths 2examination Paper

- Personalized Recommendations
 - Nmaths 2examination Paper User Reviews and Ratings
 - Nmaths 2examination Paper and Bestseller Lists
5. Accessing Nmaths 2examination Paper Free and Paid eBooks
 - Nmaths 2examination Paper Public Domain eBooks
 - Nmaths 2examination Paper eBook Subscription Services
 - Nmaths 2examination Paper Budget-Friendly Options
 6. Navigating Nmaths 2examination Paper eBook Formats
 - ePub, PDF, MOBI, and More
 - Nmaths 2examination Paper Compatibility with Devices
 - Nmaths 2examination Paper Enhanced eBook Features
 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Nmaths 2examination Paper
 - Highlighting and Note-Taking Nmaths 2examination Paper
 - Interactive Elements Nmaths 2examination Paper
 8. Staying Engaged with Nmaths 2examination Paper
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Nmaths 2examination Paper
 9. Balancing eBooks and Physical Books Nmaths 2examination Paper
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Nmaths 2examination Paper
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Nmaths 2examination Paper
 - Setting Reading Goals Nmaths 2examination Paper
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Nmaths 2examination Paper

- Fact-Checking eBook Content of Nmaths 2examination Paper
- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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