

Week 14: (Lesson 2, Part 1) Gases and Thermochemistry

Objectives for Week 14

By the end of this unit, students should be able to describe the properties of a gas, and the relationship between pressure, volume, and temperature. They should also be able to describe the relationship between the amount of gas and the volume it occupies. They should also be able to describe the relationship between the amount of gas and the pressure it exerts. They should also be able to describe the relationship between the amount of gas and the temperature it exerts. They should also be able to describe the relationship between the amount of gas and the volume it occupies. They should also be able to describe the relationship between the amount of gas and the pressure it exerts. They should also be able to describe the relationship between the amount of gas and the temperature it exerts.

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Course Learning Objectives

Knowledge and Understanding

1. Explain the relationship between pressure, volume, and temperature for a gas.
2. Explain the relationship between the amount of gas and the volume it occupies.
3. Explain the relationship between the amount of gas and the pressure it exerts.
4. Explain the relationship between the amount of gas and the temperature it exerts.

Related Unit Resources

Students should be able to describe the properties of a gas, and the relationship between pressure, volume, and temperature. They should also be able to describe the relationship between the amount of gas and the volume it occupies. They should also be able to describe the relationship between the amount of gas and the pressure it exerts. They should also be able to describe the relationship between the amount of gas and the temperature it exerts. They should also be able to describe the relationship between the amount of gas and the volume it occupies. They should also be able to describe the relationship between the amount of gas and the pressure it exerts. They should also be able to describe the relationship between the amount of gas and the temperature it exerts.

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