

## 10.4 Electrophilic Reactions

**General Introduction** An electrophile is an electron deficient species, reacting with a lone pair or a double bond.

### Electrophilic Substitution

Electrophilic Substitution (E<sub>1</sub> or E<sub>2</sub>) is a reaction in which an electrophile reacts with a nucleophile.

- It is a reaction in which a nucleophile (Nu<sup>-</sup>) attacks an electrophile (E<sup>+</sup>) to form a new bond.
- The nucleophile attacks the electrophile at the site of the electrophile, which is called the electrophilic site.

Electrophilic Substitution reactions are classified into two types: E<sub>1</sub> and E<sub>2</sub>.

- The first step in the E<sub>1</sub> mechanism is the loss of a leaving group from the substrate to form a carbocation intermediate.
- The second step is the attack of the nucleophile on the carbocation to form the product.

When reacting a nucleophile with an electrophile, the following reactions occur:

### Electrophilic Substitution

- In the following mechanism, a nucleophile (Nu<sup>-</sup>) attacks an electrophile (E<sup>+</sup>) to form a new bond. The nucleophile attacks the electrophile at the site of the electrophile, which is called the electrophilic site.
- Before the reaction, the nucleophile (Nu<sup>-</sup>) and the electrophile (E<sup>+</sup>) are separated.
- As the nucleophile attacks the electrophile, a new bond is formed between them. This is called the transition state.
- After the reaction, the nucleophile (Nu<sup>-</sup>) and the electrophile (E<sup>+</sup>) are separated.
- The reaction is complete when the nucleophile (Nu<sup>-</sup>) and the electrophile (E<sup>+</sup>) are separated.

### Transition

- When a nucleophile attacks an electrophile, a transition state is formed.

# Reading And Study Workbook 19 4 Neutralization Reactions

**B Lingard**



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