

B-3. Dissolution Rates: Practice Problems

1. Consider the reaction of the gasification of coal: (4) 2, 100 pts.



2. Consider the gasification of a powdered sample of coal in a reactor at 1000°C. How much carbon is consumed per mole of H_2 gas produced?

2. The unbalanced reaction for the steam-gasification of bituminous coal is shown (20 points) (balanced in solution, below):



a. (10 pts) Write the most basic equation for this reaction.

- b. This is not a valid equation. Explain why. Write the balanced equation. (10 pts) (Note: If you do not balance the equation, you will be penalized in the final assessment. Indicate if the unbalanced equation does not meet stoichiometry.)

c. These reactions are balanced but you are told the correct coefficients. Please write steps to show how you arrived at the correct equation. Your answer should be shown.

3. The reaction is given below (10 points):

Reaction Rates And Equilibrium Practice Problems Answer

S Ashworth



Reaction Rates And Equilibrium Practice Problems Answer:

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