

Name _____ Date _____ Class _____

CHAPTER 15 **STUDY GUIDE FOR CONTENT MASTERY**

Solutions

Section 15.1 What are solutions?

In your textbook, read about the characteristics of solutions.

Use each of the terms below just once to complete the passage.

immiscible	liquid	soluble	solution
insoluble	miscible	solvent	solvent

Air is air! (1) solution of oxygen gas dissolved in nitrogen gas. The oxygen in air is the (2) solute and nitrogen is the (3) solvent. Because oxygen gas dissolves in a solvent, oxygen gas is air's (4) soluble substance. A substance that does not dissolve is (5) insoluble. (6) Liquid solutions are the most common type of solutions. If one liquid is soluble in another liquid, such as acetic acid in water, the two liquids are (7) miscible. However, if one liquid is insoluble in another, the liquids are (8) immiscible.

Read about solvation in aqueous solutions in your textbook.

The diagram shows the hydration of solid sodium chloride to form an aqueous solution. Use the diagram to answer the following questions.



8. Hydration is solvation in which the solvent is water. What is solvation?

Solvation is the process of surrounding solute particles with solvent particles to form a solution.

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CHAPTER 15 **STUDY GUIDE FOR CONTENT MASTERY**

Section 15.1 continued

10. As sodium chloride dissolves in water, what happens to the sodium and chloride ions?

The sodium and chloride ions are separated and surrounded by the water molecules.

11. Explain the attraction of the water molecules around the sodium ions and chloride ions.

Because the sodium ion is positively charged, it attracts the negatively charged portion of the water molecule (the oxygen atom) and repels the positively charged portion of the water molecule (the hydrogen atoms). Because the chloride ion is negatively charged, it attracts the positively charged portion of the water molecule and repels the negatively charged portion.

12. How does the strength of the attraction between water molecules and sodium and chloride ions compare with the strength of the attraction between the sodium ions and chloride ions? How do you know?

The attraction between the water molecules and the sodium and chloride ions is greater than the attraction between the sodium and chloride ions. The greater strength of attraction between the water molecules and the ions is what causes the solvation process to occur.

13. List three ways that the rate of solvation may be increased.

stirring or shaking the solution, breaking the solute into smaller pieces, and heating the solvent

In your textbook, read about heat of solution, solubility, and factors that affect solubility.

For each statement below, write true or false.

true 14. The overall energy change that occurs when a solution forms is called the heat of solution.

false 15. Solubility is a measure of the maximum amount of solute that dissolves in a given amount of solvent at a specified temperature and pressure.

false 16. Solvation continues as long as the solvation rate is less than the crystallization rate.

true 17. In a saturated solution, solvation and crystallization are in equilibrium.

true 18. Additional solute can be dissolved in an unsaturated solution.

true 19. The solubility of a gas dissolved in a liquid decreases as the temperature of the solution increases.

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Reaction Spontaneity Study Guide For Content Mastery

**W. Gareth Evans, Robert Smith, Gareth
Elwyn Jones**



Reaction Spontaneity Study Guide For Content Mastery:

Resources in Education ,1984 **Your Teaching Style** Kay Mohanna,Ruth Chambers,David Wall,2018-05-08 This extraordinary and practical book examines neuro linguistic programming NLP the knowledge and skills to detect and affect thinking patterns and applies it to each phase of the medical consultation It outlines the NLP tools most useful to physicians who wish to understand and utilise the dynamic structure underlying the processes used by excellent communicators It explains how improving communication skills and developing new models of consultation to incorporate into daily practice not only helps healthcare professionals become better communicators but reassures patients alleviating suffering and promoting healing This book provides many case examples and includes skill based exercises to ensure easy and effective learning There are unique fresh perspectives on challenging areas such as anger and aggression dealing with complaints breaking bad news the heartsink patient uncovering hidden depression and telephone consulting skills It is relevant to all healthcare professionals and of special interest to general practitioners GP trainers counsellors and medical students Building on the Calgary Cambridge model Lewis Walker has outlined some of the NLP tools that are most useful to physicians who wish to engage their patients physical and psychological ability to self heal It is truly a pleasure to introduce a book that offers pragmatic tools in the service of that highest calling namely the desire to alleviate suffering and promote healing Ian McDermott in his Foreword This is a book packed with practical advice and a welcome feature is the way the author relates his message to everyday problems and concerns that patients bring to the surgery It contains powerful tools for change to be used ethically and with compassion John Duncan in his Foreword **Instructional Design for LIS Professionals** Melissa A. Wong,2019-05-03 A concise practical guide to effectively teaching current and future librarians in graduate programs professional settings and beyond Many librarians are thrust into positions where they are asked to teach colleagues Others choose to share their knowledge and experience by preparing the next generation of librarians in graduate programs However few such librarians have received any formal education in instructional design In this book Melissa A Wong an expert instructional designer helps information professionals to prepare for their roles as teachers of current and future librarians Covering topics that range from syllabus construction to evaluation and student feedback the book offers practical guidance on how to communicate with and support learners and how to come up with assignments and grade them along with advice on accessibility issues and working with technologies such as LMS OER videos and PowerPoint It demonstrates how to adapt principles of effective teaching to settings including workshops professional development courses conference presentations and staff training It also discusses professional challenges such as managing workload and shows how to adapt formal coursework to informal teaching situations Librarians who wish to learn new methods or improve on their teaching and course design skills should read this book *Proceedings* Council on Social Work Education,1957 **Methods for Effective Teaching** Paul R. Burden,David M. Byrd,2010 Methods for Effective Teachingprovides research based coverage of

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