

Name: _____ per: _____

Worksheet - Reaction Rates

Use this reaction for the questions below: $C_6H_{12}O_6 (s) + 6 O_2(g) \rightarrow 6 H_2O (g) + 6 CO_2 (g)$

- What happens to the concentrations of:
 - $C_6H_{12}O_6$ & O_2 as the reaction proceeds $\rightarrow$?
 - H_2O + CO_2 as the reaction proceeds $\rightarrow$?
- According to the collision theory, what 3 circumstances are needed for $C_6H_{12}O_6$ & O_2 to react?
- What is the activation energy for a chemical reaction?
- Use the equation & the collision theory to explain:



Change in condition:	Does this <u>increase</u> or <u>decrease</u> the rate of reaction?	Explain why
a. <u>Increasing the temperature</u>	Ex: Increase (springs up)	Ex: Molecules move faster & collide more = increased rxn rate
b. <u>Increasing the concentration</u> of $C_6H_{12}O_6$		
c. <u>Decreasing the concentration</u> of O_2		
d. <u>Increase the surface area</u> by chewing up food in your mouth.		
e. <u>Decreasing the temperature</u>		
f. <u>Increasing the pressure</u> in the container		
g. <u>Decreasing the concentration</u> of H_2O		
h. <u>Increasing the volume</u> of the container the reaction occurs in		
i. <u>Increasing the concentration</u> of CO_2		
j. <u>Using a catalyst</u> (like salivary amylase)		

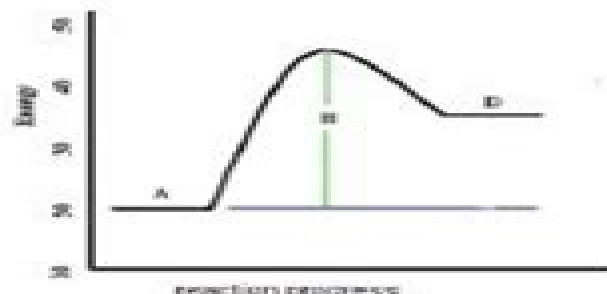
- On the accompanying energy diagram, label the following terms:

a. reactants b. products c. activation energy

- On the graph to the right, draw and label what this diagram would look like if a catalyst was added to the reaction.

- Graph reading

- How much energy (kJ) do the reactants have?
- How much energy (kJ) do the products have?
- How much energy (kJ) is required to activate this un-catalyzed reaction?
- Is this reaction endothermic or exothermic? How do you know?



Reaction Rates Equilibrium Worksheet Answers Chapter 19

Michael Brown



Reaction Rates Equilibrium Worksheet Answers Chapter 19:

Unveiling the Energy of Verbal Artistry: An Psychological Sojourn through **Reaction Rates Equilibrium Worksheet Answers Chapter 19**

In some sort of inundated with screens and the cacophony of fast communication, the profound energy and psychological resonance of verbal artistry usually diminish in to obscurity, eclipsed by the regular barrage of noise and distractions. Yet, located within the musical pages of **Reaction Rates Equilibrium Worksheet Answers Chapter 19**, a captivating work of fictional elegance that pulses with raw feelings, lies an wonderful journey waiting to be embarked upon. Written by way of a virtuoso wordsmith, that exciting opus instructions viewers on a mental odyssey, delicately exposing the latent possible and profound influence stuck within the complicated web of language. Within the heart-wrenching expanse of the evocative evaluation, we shall embark upon an introspective exploration of the book is key themes, dissect its interesting writing design, and immerse ourselves in the indelible impression it leaves upon the depths of readers souls.

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