

## Model 2 – Ternary Acids (Oxyacids)

Acid	Name of Acid in Aqueous Solution	Cation (+)	Polyatomic Anion (-)	Polyatomic Anion Name
$\text{HClO}_3$	Chloric acid	$\text{H}_3\text{O}^+$	$\text{ClO}_3^{-1}$	Chlorate
$\text{H}_2\text{SO}_3$	Sulfurous acid	$2\text{H}_3\text{O}^+$	$\text{SO}_3^{-2}$	Sulfite
$\text{H}_2\text{SO}_4$	Sulfuric acid	$2\text{H}_3\text{O}^+$	$\text{SO}_4^{-2}$	Sulfate
$\text{H}_3\text{PO}_3$	Phosphorous acid	$3\text{H}_3\text{O}^+$	$\text{PO}_3^{-3}$	Phosphite
$\text{H}_3\text{PO}_4$	Phosphoric acid	$3\text{H}_3\text{O}^+$	$\text{PO}_4^{-3}$	Phosphate
$\text{HNO}_3$	Nitric acid	$\text{H}_3\text{O}^+$	$\text{NO}_3^{-1}$	Nitrate
$\text{HNO}_2$	Nitrous acid	$\text{H}_3\text{O}^+$	$\text{NO}_2^{-1}$	Nitrite
$\text{H}_2\text{CO}_3$	Carbonic acid	$2\text{H}_3\text{O}^+$	$\text{CO}_3^{-2}$	Carbonate

6. Look at the formulas of the ternary acids in Model 2.

a. How are ternary acids different from binary acids in their structure?

*Ternary acids contain three elements and they also contain the element oxygen.*

b. What number do you think the prefix "ter-" refers to?

*"ter" refers to three*

7. When ternary acids are mixed with water, ions will form. Fill in the table above with the formulas and names of the anions.

8. Examine the pairs of ternary acids in Model 2 that contain sulfur, phosphorus, and nitrogen. Each pair has one acid that ends in "-ic" and another that ends in "-ous." These endings are related to the name of the polyatomic anion found in the acid ("-ate" or "-ite"). Complete the statements below with the correct acid name ending.

Polyatomic anion ending is "-ate" → acid name ending is "ic"

Polyatomic anion ending is "-ite" → acid name ending is "ous"

9. If the prefix "hydro-" were used to name a ternary acid, what problem would this create when naming  $\text{HClO}_3$ ?

*$\text{ClO}_3^{-1}$  has the name of Chlorate. "ate" must be replaced with "ic". If "hydro" is used as a prefix, it would be hydrochloric acid. This has the formula  $\text{HCl}$ , not  $\text{HClO}_3$ .*

10. Write a rule for naming ternary acids.

*- The prefix "hydro" is not used in the name when an acid contains more than two elements and one of the elements is oxygen.*

*- "ate" is replaced with "ic" and "ite" is replaced with "ous".*

11. Predict the formula for chlorous acid.



12. Circle the acid(s) below that would be named beginning with the prefix "hydro-."



# Polyatomic Pogil Answers

**Huangqi Zhang**



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## **Table of Contents Polyatomic Pogil Answers**

1. Understanding the eBook Polyatomic Pogil Answers
  - The Rise of Digital Reading Polyatomic Pogil Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying Polyatomic Pogil Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Polyatomic Pogil Answers
  - User-Friendly Interface
4. Exploring eBook Recommendations from Polyatomic Pogil Answers
  - Personalized Recommendations
  - Polyatomic Pogil Answers User Reviews and Ratings
  - Polyatomic Pogil Answers and Bestseller Lists
5. Accessing Polyatomic Pogil Answers Free and Paid eBooks
  - Polyatomic Pogil Answers Public Domain eBooks
  - Polyatomic Pogil Answers eBook Subscription Services
  - Polyatomic Pogil Answers Budget-Friendly Options

6. Navigating Polyatomic Pogil Answers eBook Formats
  - ePub, PDF, MOBI, and More
  - Polyatomic Pogil Answers Compatibility with Devices
  - Polyatomic Pogil Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Polyatomic Pogil Answers
  - Highlighting and Note-Taking Polyatomic Pogil Answers
  - Interactive Elements Polyatomic Pogil Answers
8. Staying Engaged with Polyatomic Pogil Answers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Polyatomic Pogil Answers
9. Balancing eBooks and Physical Books Polyatomic Pogil Answers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Polyatomic Pogil Answers
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Polyatomic Pogil Answers
  - Setting Reading Goals Polyatomic Pogil Answers
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Polyatomic Pogil Answers
  - Fact-Checking eBook Content of Polyatomic Pogil Answers
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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