

B. Answer any 2 questions from 18 to 20. Each carries 2 scores. (2 × 2 = 4)

18. Why is Cr^{2+} reducing and Mn^{3+} oxidizing when both have d^4 configuration ?
19. Explain carbylamine reaction with equation.
20. $\text{CH}_3 - \text{NH}_2$ is more basic than NH_3 while $\text{C}_6\text{H}_5 - \text{NH}_2$ is less basic than NH_3 . Explain.

PART – III

A. Answer any 3 questions from 21 to 24. Each carries 3 scores. (3 × 3 = 9)

21. Write any three differences between crystalline and amorphous solids. (3)
22. 18g of glucose, $\text{C}_6\text{H}_{12}\text{O}_6$, is dissolved in 1 kg of water in a sauce pan. At what temperature will water boil at 1.013 bar ? (3)
(K_b for water is $0.52 \text{ K kg mol}^{-1}$, boiling point of water = 373.15 K)
23. (i) Mention any two factors which influence the rate of a chemical reaction. (1)
(ii) Derive an expression for half life of a first order reaction from its integrated rate equation. (2)
24. (i) Write a suitable method to convert. (1)
 $\text{CH}_3 - \text{CH}_2 - \text{Br}$ to $\text{CH}_3 - \text{CH}_2 - \text{I}$
(ii) Suggest and explain a suitable mechanism for the nucleophilic substitution of tert – butyl bromide with NaOH. (2)

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Xiaolong Qi



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Table of Contents Plus Two Kerala Chemistry Model Question Paper

1. Understanding the eBook Plus Two Kerala Chemistry Model Question Paper
 - The Rise of Digital Reading Plus Two Kerala Chemistry Model Question Paper
 - Advantages of eBooks Over Traditional Books
2. Identifying Plus Two Kerala Chemistry Model Question Paper
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Plus Two Kerala Chemistry Model Question Paper
 - User-Friendly Interface
4. Exploring eBook Recommendations from Plus Two Kerala Chemistry Model Question Paper
 - Personalized Recommendations
 - Plus Two Kerala Chemistry Model Question Paper User Reviews and Ratings
 - Plus Two Kerala Chemistry Model Question Paper and Bestseller Lists
5. Accessing Plus Two Kerala Chemistry Model Question Paper Free and Paid eBooks
 - Plus Two Kerala Chemistry Model Question Paper Public Domain eBooks
 - Plus Two Kerala Chemistry Model Question Paper eBook Subscription Services
 - Plus Two Kerala Chemistry Model Question Paper Budget-Friendly Options

6. Navigating Plus Two Kerala Chemistry Model Question Paper eBook Formats
 - ePub, PDF, MOBI, and More
 - Plus Two Kerala Chemistry Model Question Paper Compatibility with Devices
 - Plus Two Kerala Chemistry Model Question Paper Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Plus Two Kerala Chemistry Model Question Paper
 - Highlighting and Note-Taking Plus Two Kerala Chemistry Model Question Paper
 - Interactive Elements Plus Two Kerala Chemistry Model Question Paper
8. Staying Engaged with Plus Two Kerala Chemistry Model Question Paper
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Plus Two Kerala Chemistry Model Question Paper
9. Balancing eBooks and Physical Books Plus Two Kerala Chemistry Model Question Paper
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Plus Two Kerala Chemistry Model Question Paper
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Plus Two Kerala Chemistry Model Question Paper
 - Setting Reading Goals Plus Two Kerala Chemistry Model Question Paper
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Plus Two Kerala Chemistry Model Question Paper
 - Fact-Checking eBook Content of Plus Two Kerala Chemistry Model Question Paper
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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