BALANCING CHEMICAL EQUATIONS

```
Married
Directions: Balance each equation by writing a coefficient in each blank
           If a coefficient is equal to 1, you do not have to write it.
    _Cu_PO_(s) + __NH_J(aq) -> __Cul(aq) + __(NH_J)PO_(aq)
    __KOH(aq) + ___ L(h) --- __ LIOH(aq) + ___ K(h)
    Fe(OH)_(s) + ___ Na(s) -> __ NaOH(as) + ___ Fe(s)
     CH(OH(g) + ___O(g) --- ___CO.(g) + ___H(O(g)
    NaHCO(s) -- Na,CO(s) - CO(g) - H,O(l)
     Sr(OH),(aq) + ___ Na,PO,(aq) -> ___ NaOH(aq) + ___ Sr,(PO,),(x)
     Al(s) + ___ Fe(NO<sub>3</sub>), (44) -- ___ Fe(s) + ___ Al(NO<sub>3</sub>),(44)
     C_H_(O_(s) + ___O_(g) -> ___CO_(g) + ___H_(O(g)
      Mn(SO,),(aq) + ___ Na,O(s) -- ___ MnO,(s) + ___ Na,SO,(aq)
    ____CaCOJN --- ___ CNON1+ ___ COJN
     __Al,(CrO<sub>2</sub>), (aq) + ___ Mgl,(aq) -> ___ MgCrO<sub>2</sub>(aq) + ___ All,(aq)
    _____C,H,N,O,(s) --- _____N,(g) + _____H,(g) + _____CO(g) + ____C(s)
     Re(O_i(s) = Re(s) \rightarrow ReO_i(s)
    P_(x) = N_(O(x) -> P_(O(x) + ____ N_(x)
     _ MnO;(s) + __ HO(aq) --- __ MnO;(aq) + __ O;(g) + __ H;O(t)
     __Na<sub>2</sub>S<sub>2</sub>O<sub>2</sub>(eq) = ___1(eq) -> ___Nai(eq) • ___Na<sub>2</sub>S<sub>2</sub>O<sub>2</sub>(eq)
       ALC(6) + __H,O(3 -> __ A(OH)(10) + __ OH(6)
      NICU(as) + ____ Na,PO_(as) --- ___ Ni,(PO_)(s) + __
```

A CHEMISTRY WORKSHEET

20 PROBLEMS

WORKED ANSWER KEY INCLUDED

Plato Learning Answer Key Chemistry

Michael Seilmaier

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