

*Rock n' Blues*

# HARMONICA



**A Beginner's Guide  
to Jamming**

**JON GINDICK**

# Rock N Blues Harmonica A Beginner S Guide To Jamming

**JL Elias**



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web for example take the reaction of lead ii nitrate with hydrochloric acid to form lead ii chloride and nitric acid shown below  
$$\text{Pb(NO}_3)_2 \text{ (aq)} + 2 \text{HCl (aq)} \rightarrow \text{PbCl}_2 \text{ (s)} + 2 \text{HNO}_3$$

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web answer key to practice problems on net ionic equations molecular  $\text{AgNO}_3$  (aq)  $\text{KCl}$  (aq)  $\text{AgCl}$  (s)  $\text{KNO}_3$  (aq) total ionic  $\text{Ag}^+$  (aq)  $\text{NO}_3^-$  (aq)  $\text{K}^+$  (aq)  $\text{Cl}^-$  (aq)  $\text{AgCl}$  (s)

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web you may have to consult the solubility rules write the net ionic equation for the reaction of  $\text{BaCl}_2(\text{aq})$  and  $\text{Na}_2\text{SO}_4(\text{aq})$  you may have to consult the solubility rules write

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web choose the correct net ionic equation representing the reaction that occurs when solutions of potassium carbonate and copper i chloride are mixed a  $\text{K}_2\text{CO}_3(\text{aq}) + 2\text{CuCl}(\text{aq})$

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web the net ionic equation for equation 1 is given as equation 3  $\text{s}^{2+}(\text{aq}) + \text{Fe}^{2+}(\text{aq}) \rightarrow \text{Fe}(\text{s}) + \text{S}(\text{s})$  eq 3 8 note that the phase indicator aq is optional on ionic species but is

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web  $\text{CuClO}_4$  is soluble all perchlorates are soluble while  $\text{PbSO}_4$  is insoluble sulfates are generally soluble except for  $\text{Pb}^{2+}$   $\text{Ba}^{2+}$   $\text{Hg}_2^{2+}$  and  $\text{Sr}^{2+}$  the net ionic equation is

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web balanced chemical equation  $2\text{NaOH}(\text{aq}) + \text{MgCl}_2(\text{aq}) \rightarrow 2\text{NaCl}(\text{aq}) + \text{Mg}(\text{OH})_2(\text{s})$  full ionic equation  $2\text{Na}^+(\text{aq}) + 2\text{OH}^-(\text{aq}) + \text{Mg}^{2+}(\text{aq}) + 2\text{Cl}^-(\text{aq}) \rightarrow 2\text{Na}^+(\text{aq}) + 2\text{Cl}^-(\text{aq}) + \text{Mg}(\text{OH})_2(\text{s})$

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