

Reaction Energy

Question 1: Thermodynamics

In the space provided, write the letter of the term or phrase that best completes each sentence or answers each question.

- _____ 1. The sign convention for the study of thermodynamics is that:
a. energy + changes during heating.
b. energy + changes during + bonds dissociation.
c. a rise - changes during chemical reactions.
d. heat is endothermic.
- _____ 2. Energy is measured in units of:
a. joules.
b. calories.
c. grams.
d. grams.
- _____ 3. Heat gained is used to measure enthalpy?
a. H.
b. ΔH
c. Δ
d. None of the above.
- _____ 4. The specific heat of a substance is the amount of energy needed to
raise the temperature of
a. 1 g by 1 K.
b. 1 mol by 1 g.
c. 1 g by 1 K.
d. 1 mol by 1 K.
- _____ 5. Effects of the following on the heat of combustion (the energy) required to
burn the temperature of a substance:
a. decrease of the substance.
b. the specific heat of the substance.
c. the mass of the substance.
d. the temperature of the substance.

Reaction Energy Section Quiz Modern Chemistry

University of California, Los Angeles

A red circular graphic with a gradient, appearing as a partial circle or a stylized 'C' shape, located to the right of the University of California, Los Angeles text.

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