Name: Class:_ Five Types of Chemical Reaction Worksheet Answers Balance the following reactions and indicate which of the six types of chemical reaction are being 2 NaBr + 1 Ca(OH)₂ → 1 CaBr₂ + 2 NaOH Type of reaction: double displacement að. $2 NH_1 + 1 H_2SO_4 \rightarrow 1 (NH_4)_2SO_4$ **b**) Type of reaction: synthesis 4 C₅H₉O + 27 O₂ → 20 CO₂ + 18 H₂O Type of reaction: combustion CB. di) 3 Pb + 2 H₂PO₄ → 3 H₂ + 1 Pb₁(PO₄)₂ Type of reaction: single displacement Type of reaction: double displacement et. 1 Li₂N + 3 NH₄NO₃ → 3 LiNO₃ + 1 (NH₄)₃N Section 3: Predicting the products of chemical reactions 1) 2 Ag + 1 CuSO₄ → 1 Ag₂SO₄ + 1 Cu Type: Single Displacement 23 2 Nai + 1 CaCl₂ → 2 NaCl + 1 Cal₂ Type: Double Displacement 30 1 O2 + 1 H2 → 2 H2O Type: Synthesis 400 2 AgNO₂ + 1 BaSO₄ → 1 Ag₂SO₄ + 1 Ba(NO₂)₂ Type: Double Displacement 51 2 HCN + 1 CuSO₄ → 1 H₂SO₄ + 1 Cu(CN)₂ Type: Double Displacement 6) 1 H₂O + 1 AgI → 1 HI + 1 AgOH Type: Double Displacement 73 4 LiBr + 1 Co(SO₁)₂ → 2 Li₂SO₁ + 1 CoBr₄ Type: Double Displacement

Type: Single Displacement

Type: Single Displacement

Type: Single Displacement

Type: Double Displacement

Type: Single Displacement

Type: Double Displacement

Type: Decomposition

Type: Synthesis

Type: Decomposition

Type: Combustion

Type: Combustion

8):

9)

1.0%

1.13

12)

13)

14)

151

1.6)

171

18)

1 LiNO₁ + 1 Ag → 1 AgNO₁ + 1 Li

1 N2 + 2 O2 → 2 NO2

1 H₂CO₃ → 1 CO₂ + 1 H₂O

1 AlCl₃ + 3 Cs → 3 CsCl + 1 Al

1 C₀H₀ + 6 O₂ → 4 CO₂ + 4 H₂O

1 BaS + 1 PtCl₂ → 1 BaCl₂ + 1 PtS

2 Na₂O → 4 Na + 1 O₂

1 Al(NO₃)₃ + 1 Ga → 1 Ga(NO₃)₃ + 1 Al

2 KCl + 1 Mg(OH)₂ → 2 KOH + 1 MgCl₂

1 Zn + 1 Au(NO₂)₂ → 1 Zn(NO₂)₂ + 1 Au

1 CH₃COOH + 2 O₂ → 2 CO₂ + 2 H₂O

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- 1. Understanding the eBook Reaction Types Answers Chemistry
 - The Rise of Digital Reading Reaction Types Answers Chemistry
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Reaction Types Answers Chemistry
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Reaction Types Answers Chemistry
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Reaction Types Answers Chemistry
 - Personalized Recommendations
 - Reaction Types Answers Chemistry User Reviews and Ratings
 - Reaction Types Answers Chemistry and Bestseller Lists
- 5. Accessing Reaction Types Answers Chemistry Free and Paid eBooks

- Reaction Types Answers Chemistry Public Domain eBooks
- Reaction Types Answers Chemistry eBook Subscription Services
- Reaction Types Answers Chemistry Budget-Friendly Options
- 6. Navigating Reaction Types Answers Chemistry eBook Formats
 - o ePub, PDF, MOBI, and More
 - Reaction Types Answers Chemistry Compatibility with Devices
 - Reaction Types Answers Chemistry Enhanced eBook Features
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 - Highlighting and Note-Taking Reaction Types Answers Chemistry
 - Interactive Elements Reaction Types Answers Chemistry
- 8. Staying Engaged with Reaction Types Answers Chemistry
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Reaction Types Answers Chemistry
- 9. Balancing eBooks and Physical Books Reaction Types Answers Chemistry
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Reaction Types Answers Chemistry
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Reaction Types Answers Chemistry
 - Setting Reading Goals Reaction Types Answers Chemistry
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Reaction Types Answers Chemistry
 - Fact-Checking eBook Content of Reaction Types Answers Chemistry
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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