# TYPES OF CHEMICAL REACTIONS

Identify each type of reaction:

- 1. Na<sub>3</sub> (PO<sub>4</sub>) + 3 KOH --> 3 NaOH + K<sub>3</sub> (PO<sub>4</sub>) double displacement
- MgCl<sub>2</sub> + Li<sub>2</sub>(CO<sub>3</sub>) --> Mg(CO<sub>3</sub>) + 2 LiCl double displacement
- 3. C<sub>6</sub>H<sub>12</sub> + 9 O<sub>2</sub> --> 6 CO<sub>2</sub> + 6 H<sub>2</sub>O combustion
- 4. Pb + Fe(SO<sub>4</sub>) --> Pb(SO<sub>4</sub>) + Fe single displacement
- 5. 2 AgNO<sub>3</sub> + Cu --> Cu(NO<sub>3</sub>)<sub>2</sub> + 2 Ag single displacement
- 6. C<sub>3</sub>H<sub>6</sub>O + 4 O<sub>2</sub> --> 3 CO<sub>2</sub> + 3 H<sub>2</sub>O combustion
- 2 (C<sub>5</sub>H<sub>5</sub>) + Fe --> Fe(C<sub>5</sub>H<sub>5</sub>)<sub>2</sub>
  synthesis
- 8. CaCO<sub>3</sub> --> CaO + CO<sub>2</sub> decomposition
- 9. P<sub>4</sub> + 3 O<sub>2</sub> --> 2 P<sub>2</sub>O<sub>3</sub> synthesis
- 10. 2 RbNO<sub>3</sub> + BeF<sub>2</sub> --> Be(NO<sub>3</sub>)<sub>2</sub> + 2 RbF double displacement

# **Reaction Products Worksheet Answers**

**S Baum** 

**Reaction Products Worksheet Answers:** 

This book delves into Reaction Products Worksheet Answers. Reaction Products Worksheet Answers is an essential topic that must be grasped by everyone, from students and scholars to the general public. The book will furnish comprehensive and indepth insights into Reaction Products Worksheet Answers, encompassing both the fundamentals and more intricate discussions.

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