

Blue cards participate _____ of the possible _____ of atoms _____
 1. _____
 2. _____

Atomic number number _____ of protons _____ in the nucleus _____ of an atom _____

Electron shells all of protons _____ $(+)$ = all of electrons _____ $(-)$

Isotopes atoms _____ of an element _____ that have different _____ numbers _____ of neutrons _____

Isotopes of Hydrogen

Hydrogen-1

1. _____ protons and _____ neutrons

Hydrogen-2

1. _____ protons and _____ neutrons

Hydrogen-3

1. _____ protons and _____ neutrons

Mass number total _____ number of protons _____ and neutrons _____ in an atom _____

Example: Carbon-12 _____ 12C _____ 12C-12

| isotope | symbol | atomic | element |
|------------|----------------|--------|------------|
| Hydrogen | ^1_1H | 1 atom | Hydrogen |
| Deuterium | ^2_1H | 1 atom | Hydrogen |
| Hydrogen-3 | ^3_1H | 1 | Hydrogen-3 |

Atomic mass average _____ of the masses _____ of all the isotopes of _____

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